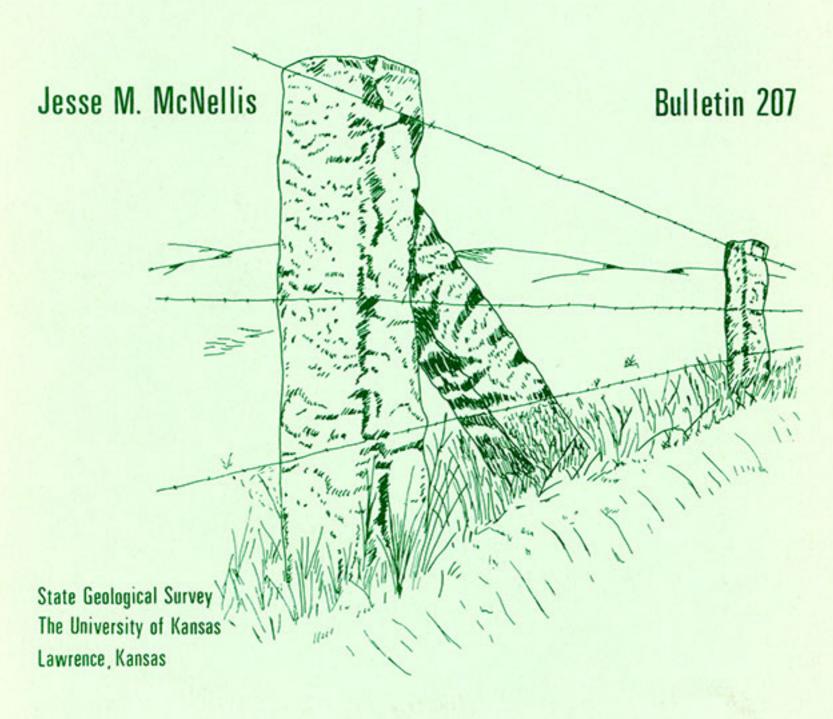
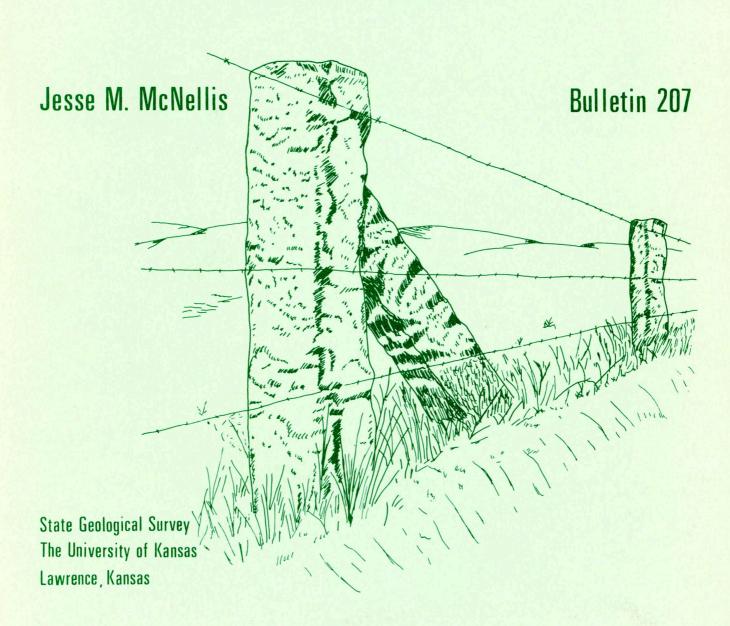
Geology and Ground-Water Resources of RUSH COUNTY Central Kansas



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BULLETIN 207

Geology and Ground-Water Resources of Rush County, Central Kansas

By Jesse M. McNellis

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Geology and Ground-Water Resources of Rush County, Central Kansas

ABSTRACT

Rush County, which comprises an area of 724 square miles in central Kansas, is principally in the Smoky Hills section of the Great Plains physiographic province. The north third of the county is in the Smoky Hill River drainage basin and the rest is in the Arkansas River drainage basin. The average annual precipitation at Bison is 22.21 inches, and the average growing season is 174 days. The population of the county in 1970 was 5,117, of which 36.6 percent was rural. Agriculture is the dominant economic activity in the county and the chief mineral resources are ground water, oil and gas, and helium.

All rocks that crop out in Rush County are sedimentary. They include the Greenhorn Limestone, Carlile Shale, and Niobrara Chalk of Cretaceous age; the Ogallala Formation of Pliocene age; and eolian, fluvial, and colluvial deposits of Pleistocene age.

The principal aquifers are the Dakota Formation, which underlies the entire county, and the Pleistocene deposits in the stream valleys. Yields from wells in the Dakota generally are adequate for stock and domestic purposes, but the chemical quality of the water may be considered marginal for most uses. Wells obtaining water from the Dakota range in depth from 80 feet in the southeastern part of the county to a reported 530 feet in the northwestern part.

The Pleistocene aquifer in Walnut Creek valley yields 290 to 1,200 gallons per minute to wells for irrigation. Yields as great as 1,500 gallons per minute have been reported. Pleistocene deposits in other stream valleys yield adequate supplies for stock and domestic purposes. Water from Pleistocene deposits commonly is hard to very hard, but is satisfactory for most uses.

Recharge to the aquifers is principally from local precipitation. The aquifer in Walnut Creek valley receives sizeable amounts of recharge during periods of high flow in Walnut Creek; perhaps as much as 1½ feet of water per acre was recharged from a flood in 1959.

INTRODUCTION

Purpose and Scope

Irrigation by wells in Walnut Creek valley in Rush County during the dry 1950's demonstrated the economic importance of the ground-water resource. Because water supplies in the county are obtained principally from wells, ground water must be utilized in the best possible manner to ensure availability of this most important natural resource for future use.

This investigation was begun in August 1959 to determine the occurrence, availability, chemical characteristics, and geologic controls of ground water in Rush County. The study is part of a continuing program of geology and ground-water-resources investigations that was begun in Kansas in 1937 by the U.S. Geological Survey and the State Geological Survey of Kansas, in cooperation with the Division of Sanitation of the Kansas State Board of Health (now the Division of Environmental Health of the Kansas State Department of Health) and the Division of Water Resources of the Kansas State Board of Agriculture. The present status of the program is shown on figure 1.

Location and Extent of Area

Rush County, in central Kansas, is in the fourth tier of counties south of the Kansas-Nebraska border and is the sixth county east of the Kansas-Colorado border (fig. 1). The bordering counties are Ellis on the north, Barton on the east, Pawnee on the south, and Ness on the west. The county contains 20 town-

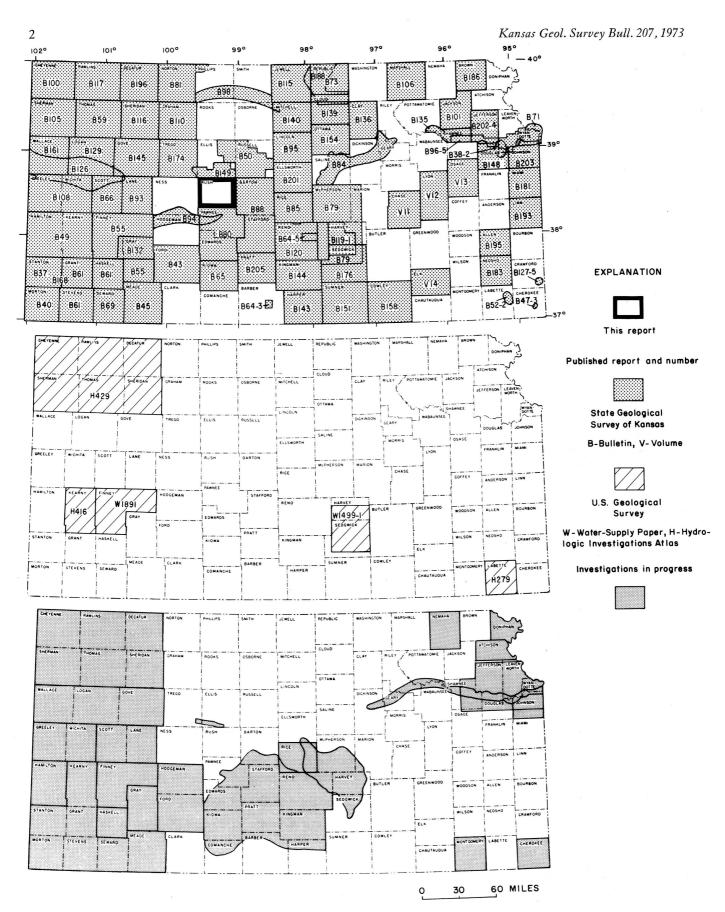


Figure 1.—Index maps showing area discussed in this report, and other areas for which ground-water reports have been published or are in preparation.

ships from T.16 S. to T.19 S. and from R.16 W. to R.20 W., and has an area of 724 square miles.

Previous Investigations

Rush County and the surrounding area have been investigated and studied previously by several workers. Reports by these investigators are listed in the Selected References.

Methods of Investigation

Field work for this report was done in the fall of 1959, the summer of 1960, and a short time in the spring of 1961. Data collected at 222 wells included depth of the well and depth to water in the well, if measurable. Other pertinent data available were obtained from well owners. A total of 161 test holes were drilled in the county by the U.S. Geological Survey and the State Geological Survey of Kansas. The test holes were drilled with a hydraulic rotary drilling machine (24 holes) and with a pickup-mounted power auger (137 holes). Drill cuttings were collected and examined. Water-well drillers in the area furnished test-hole logs. Several hundred oil-well logs on file at the State Geological Survey of Kansas were studied. In addition, a few thousand shot-hole logs from several oil and seismic survey companies were obtained and examined.

Wells and test holes were located by aerial photographs and automobile odometer readings. The altitudes of inventoried wells and Survey-drilled test holes were determined by plane table and alidade.

Samples of water were collected by the author and analyzed in the laboratory of the Division of Environmental Health of the Kansas State Department of Health. Chemical analyses of water from some of the municipal wells also were obtained from the Department of Health.

Water levels in available irrigation wells have been measured annually at least six times since 1960. Monthly or quarterly measurements are continuing on eight observation wells at the present time (1973).

Geology was mapped on aerial photographs obtained from the U.S. Department of Agriculture and transferred to a base map compiled from maps of the U.S. Soil Conservation Service and State Highway Commission of Kansas.

Well-Numbering System

Well and test-hole numbers in this report describe the location of wells and test holes according to General Land Office survey procedure as follows: first number, township; second number, range; third num-

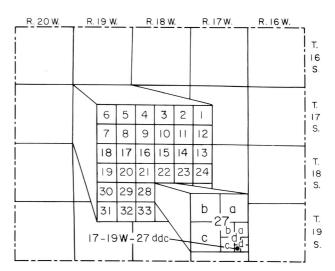


FIGURE 2.—Well-numbering system used in this report.

ber, section; first letter, 160-acre tract (quarter section) within that section; second letter, 40-acre tract (quarter-quarter section) within that quarter section; third letter, 10-acre tract (quarter-quarter-quarter section) within that quarter-quarter section. The 160-acre, 40-acre, and 10-acre tracts are designated a, b, c, and d in a counterclockwise direction beginning in the northeast corner. For example, well 17-19W-27ddc is in the SW\%SE\%SE\% sec. 27, T.17 S., R.19 W. (fig. 2). When two or more wells are located in the same 10-acre tract, the wells are numbered serially in the order they were inventoried.

Acknowledgments

Thanks and appreciation are expressed to the many county residents who permitted access to their property and supplied information on their wells, to the municipal officials who provided information on city water supplies, to well drillers who supplied well and test-hole logs, and to oil and service companies that provided shot-hole logs. Special acknowledgment is made to the owners of the irrigation wells that were used in aquifer tests, and to Mr. Jerry Oborny for the copy of water-level measurements of several years duration for his irrigation well.

Topography and Drainage

Most of Rush County is in the Smoky Hills section of the Great Plains physiographic province (Frye and Schoewe, 1953). The southeast corner of the county is in the Great Bend Region, and the northwest corner is in the High Plains.

The altitude of land surface in Rush County ranges from slightly more than 2,300 feet above mean sea level in the northwest and southwest corners of the county to slightly less than 1,850 feet in the northeast corner. The maximum relief is somewhat more than 450 feet, but local relief seldom exceeds 100 feet.

Approximately the north third of Rush County is in Smoky Hill River drainage basin, but the Smoky Hill River occurs in Rush County only in secs. 5 and 6, T.16 S., R.17 W. Big Timber Creek is the largest tributary to the Smoky Hill River in Rush County; it heads in northeastern Ness County, enters Rush County in the vicinity of McCracken, and enters Ellis County northeast of Liebenthal. Other Smoky Hill tributaries in Rush County include Shelter Creek, Duck Creek, and Eagle Creek.

The south two-thirds of Rush County is in Arkansas River drainage basin. The major stream in this part of the county is Walnut Creek, which heads in western Lane County about 55 miles west of where it enters Rush County near Alexander. Walnut Creek flows eastward across Rush County and enters Barton County east of Shaffer. Major tributaries to Walnut Creek from the south include Old Maid Fork, Sandy Creek, and Otter Creek. Alexander Dry Creek and Sand Creek are the major tributaries to Walnut Creek from the north. Dry Walnut and Dry Creeks trend east-northeast in the southeast quarter of Rush County and enter Walnut Creek in Barton County. Along the south side of Rush County are headwater areas for some tributaries of Pawnee River.

Walnut Creek is an underfit stream; that is, it appears to be too small to have eroded the valley in which it flows. Walnut Creek valley ranges from about 1 to 3 miles in width across Rush County. Prior to the development of extensive irrigation from wells in the valley-fill deposits, Walnut Creek was a perennial stream through Rush County except after long droughts.

Climate

The climate of Rush County is characterized by abundant sunshine, moderate precipitation, high rates of evaporation, moderate to high wind velocities, and frequent and abrupt weather changes. Hot days and cool nights are typical in summer. During winter, temperatures are moderate to cold with occasional short periods of severe cold. Thundershowers are prevalent in spring and summer, and blizzards occur in winter.

The first precipitation records of the National Weather Service (formerly U.S. Weather Bureau) for Rush County date from 1891 to 1893. Precipitation records resume in 1902 and, with temperature records, continue to the present. The weather station was at La Crosse until 1923, when it was moved to Bison.

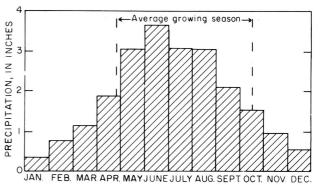


FIGURE 3.—Mean monthly precipitation 1891-93, 1902-70 and average growing season at Bison.

Three additional gages were installed at Alexander, Loretta, and McCracken in late 1940. Mean monthly precipitation through 1970 at Bison is shown on figure 3. Figure 4 shows the annual precipitation, the mean annual precipitation, and the cumulative departure from mean annual precipitation for the period of record at Bison (includes early record at La Crosse). Recorded annual precipitation in Rush County has ranged from 37.37 inches at Alexander in 1944 to 10.76 inches at Alexander in 1956.

Approximately three-quarters of the precipitation in Rush County occurs from April through September, which coincides with the growing season. The growing season at Bison has ranged from 140 to 207 days and averages 174 days.

The latest recorded date of a killing frost in the spring is May 27, 1907, and the average date is April 25. The earliest recorded date of a killing frost in the fall is September 17, 1903, and the average date is October 16. Recorded temperatures in Rush County have ranged from 116°F on July 13, 1934, to -25°F on January 12, 1912.

Population

The first settler in Rush County arrived in 1870 and the first family in 1871. Rush County was organized in 1874, and in 1875 counted 451 residents. In 1970 the county had a population of 5,117 (U.S. Bureau of the Census, 1971), of which 3,246 or 63.4 percent lived in incorporated cities and 1,871 or 36.6 percent lived on farms or in unincorporated communities. The population decreased 16.9 percent between 1960 and 1970. La Crosse, the largest city and the county seat, had a population of 1,583 in 1970. Other communities and their 1970 population were: McCracken, 333; Otis, 387; Bison, 285; Rush Center, 237; Liebenthal, 169; Alexander, 129; and Timken, 123. The average density of population in Rush County in 1970 was 7.1 per square mile as compared with 27.5 for the entire State.

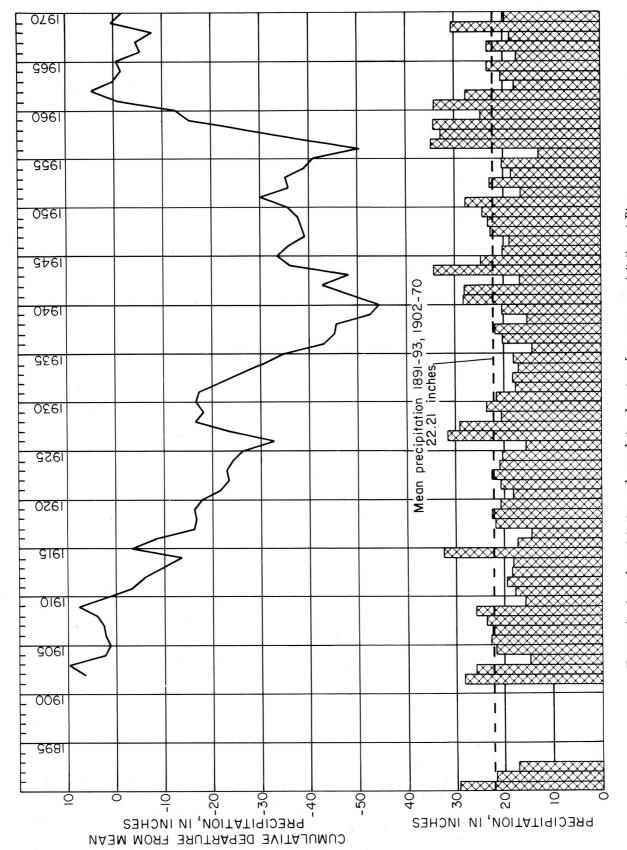


FIGURE 4.—Annual precipitation and cumulative departure from mean precipitation at Bison.

Agriculture

Agriculture is the dominant economic activity in Rush County. Crops valued at \$6,996,960 and poultry and livestock valued at \$5,017,220 were produced in Rush County in 1970 (Kansas State Board of Agriculture, 1971). The acreage and value of principal crops harvested in the county in 1970 are given in table 1.

Table 1.—Acreage and value of principal crops harvested in 1970 (Kansas State Board of Agriculture, 1971).

Crop	Acres	Value
Wheat	143,000	\$5,300,000
Sorghum, all purposes	24,800	962,300
Corn, all purposes	2,900	333,100
Oats	280	6,300
Barley	360	8,900
hay, all purposes	7,300	370,800
Other crops	210	15,560
Total	178,850	\$6,996,960

Mineral Resources

The mineral resources of Rush County include oil and gas, helium, sand and gravel, "quartzite," and limestone, in addition to ground water. The U.S. Bureau of Mines (1971) lists petroleum products, helium, and sand and gravel, having a value of \$6,749,354, as minerals produced commercially in Rush County in 1969.

OIL AND GAS

The first recorded drilling for petroleum products in Rush County was in 1903 when a dry hole in the southeast corner of sec. 33, T.17 S., R.17 W., reached a depth of 2,004 feet and another dry hole in the southwest corner of NW¼ sec. 34, T.17 S., R.17 W., reached 2,095 feet (Landes and Keroher, 1938). A third unsuccessful attempt was made in 1924 in the SW¼ sec. 3, T.16 S., R.16 W. In November 1928 the Danciger Oil and Refining Co. completed a gas well having a flow potential of 13 million cubic feet in the center of the NW¼ sec. 27, T.17 S., R.17 W., which was the beginning of successful petroleum operations in Rush County. The first oil well was completed on September 2, 1931, in the SW¼ sec. 17, T.19 S., R.16 W.

The most important producing field, the Otis-Albert gas and oil field, dates from March 1930 when the discovery gas well in sec. 11, T.18 S., R.16 W., was completed. The first oil in the field was discovered in 1934. Other important fields are the Ryan, discovered in 1945, the Rush Center, discovered in 1947, the Reichel, principally gas, discovered in 1953, and the Webs, discovered in 1954. Cumulative oil production in Rush County was 15,759,679 barrels through 1970. In addition to natural gas, helium gas has been ex-

tracted by a plant at Otis from 1943 to 1945 and from 1951 to the present time. The cumulative production, the number of wells, the producing zones, and other statistical data for all fields in the county are presented in annual reports by the Mineral Resources Section of the State Geological Survey of Kansas.

SAND AND GRAVEL

Deposits of sand and gravel are not extensive in Rush County. Remnants of Ogallala Formation provide sand and gravel in T.17 S., R.19 W., and in T.19 S., Rs.17-20 W. Pleistocene deposits in T. 16 S., Rs.16-18 W., have been used for road gravel, and a deposit of sand and gravel in sec. 3, T.18 S., R.20 W., was the source of part of the material used to build the asphalt-surfaced road between Alexander and McCracken. Small deposits of sand and gravel are present along stream courses within the county.

"QUARTZITE"

A silicified deposit of Ogallala crops out in the SW¼ sec. 6 and the NW¼ sec. 7, T. 16 S., R.20 W., and extends westward into Ness County. The quantity of material in the Rush County deposits is not large, but the material could have local use as aggregate or building stone.

LIMESTONE

The Fence-post limestone bed at the top of the Pfeifer Shale Member of the Greenhorn Limestone has been quarried extensively along the line of outcrop. No quarries are presently operating but reserves are very large. The limestone was and is used for fence-posts, as the name implies, over a wide area of central and north-central Kansas. In addition to fenceposts, the limestone has been used for foundations and buildings. The Catholic Church at Liebenthal is a fine example. The stone may be crushed and used as aggregate and for road material. Other limestone beds in the Greenhorn Limestone, and in the lowest part of the Carlile Shale, also have been quarried.

GENERAL GEOLOGY¹

Distribution of Subsurface Rocks

PRECAMBRIAN ROCKS

Rocks of Precambrian age do not crop out in Kansas. Precambrian rocks in the State have been studied using well samples, logs, and geophysical methods

¹ The classification and nomenclature of rock units used in this report are those of the State Geological Survey of Kansas and differ somewhat from those used by the U.S. Geological Survey.

(Farquhar, 1957). The recorded number of wells penetrating Precambrian rocks in Rush County through 1964 totals 224 (Cole and others, 1961, 1965). The predominant types of Precambrian rocks reported from about 70 percent of the wells were granite or granite wash. Other Precambrian rock types include quartzite, quartz, porphyry, gneiss, schist, and diabase. A map of the configuration of the top of Precambrian rocks in Kansas by Cole (1962) shows an upthrown block of Precambrian rocks (fig. 5) trending southeastward from southwest Ellis County, across Rush County, to southwest Barton County. The depth to the top of the fault block ranges from 3,500 to 3,800 feet below land surface in Rush County.

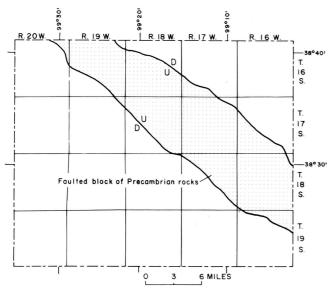


Figure 5.—Trend of faulted block of Precambrian rocks (from Cole, 1962).

Of probable interest are potassium-argon determinations made by J. Lawrence Kulp of Lamont Geological Observatory at Columbia University (Cole and Merriam, 1962) of the age of Precambrian rock samples from two wells in the county. Granite gneiss from the Skelly No. 1 Betz well in the NE¼SE¼SW¼ sec. 5, T.18 S., R.16 W., was dated at 1,260 million years and gray quartzite from the Skelly No. 4 Dyer in the NE¼SE¼NW¼ sec. 21, T.18 S., R.16 W., was dated at 1,200 million years.

PALEOZOIC ROCKS

Rocks of Paleozoic age do not crop out in Rush County. They have been studied using geophysical methods and well samples. The following maps issued by the State Geological Survey of Kansas reveal the general distribution and regional structure of Paleozoic rocks in Kansas: Arbuckle (Cambrian-Ordovician) rocks (Merriam and Smith, 1961), Silurian and De-

vonian rocks (Merriam and Kelly, 1960), Mississippian rocks (Merriam, 1960), and the Lansing Group of Pennsylvanian age (Merriam, Winchell, and Atkinson, 1958).

Briefly, distribution of Paleozoic rocks in Rush County is: Cambrian rocks are locally absent over the previously discussed upthrown Precambrian fault block (fig. 5) but are present beneath the rest of the county; Ordovician rocks are absent along the same general trend as the upthrown Precambrian fault block but are present elsewhere in the county; Silurian and Devonian rocks are absent throughout the county; Mississippian rocks are present in the southwest part of the county and absent elsewhere; Pennsylvanian and Permian rocks underlie the entire county.

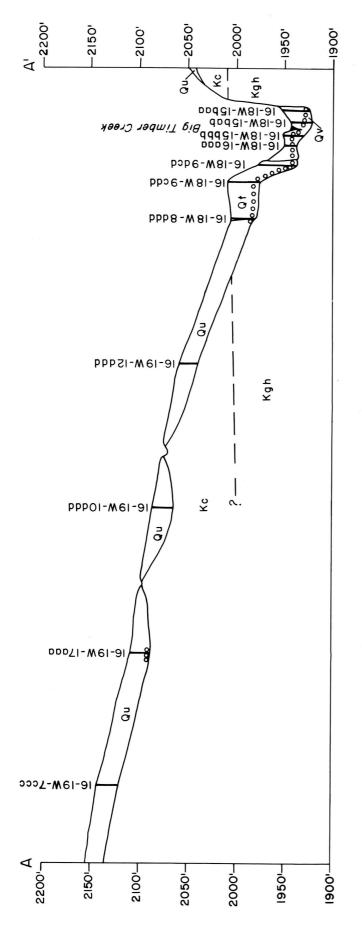
MESOZOIC ROCKS

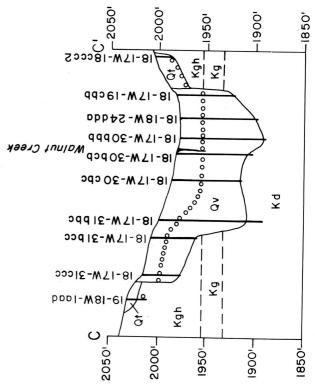
Rocks of Mesozoic age in Rush County belong to the Cretaceous System. Cretaceous rocks unconformably overlie Paleozoic rocks and, in much of the county, are overlain unconformably by Cenozoic rocks. The Cretaceous units that underline the county but do not crop out include, in ascending order: Cheyenne Sandstone, Kiowa Formation, Dakota Formation, and Graneros Shale. These units are discussed in the section on geology and water supply.

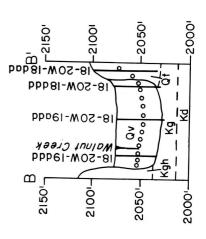
Distribution of Exposed Rocks

Rock units exposed at the land surface, as shown on the geohydrologic map of Rush County (pl. 1), are sedimentary and range in age from Cretaceous to Pleistocene. Geologic relationships and the position of the water table are shown in selected geologic sections on figure 6.

The oldest rocks exposed in Rush County are marine deposits of Late Cretaceous age. The Greenhorn Limestone is the prominent rock cropping out along most stream valleys in all but the northwest quarter and the southwest eighth of the county. The Greenhorn underlies all but the deep valleys of Walnut Creek, the Smoky Hill River, and tributaries to Walnut Creek and the Smoky Hill River. The Graneros Shale or Dakota Formation underlies these valleys. Conformably overlying the Greenhorn is the Carlile Shale (Hattin, 1962, p. 24). The lower 20 to 35 feet of Carlile is relatively resistant; it crops out in stream valleys and closely resembles Greenhorn. The Carlile Shale is the bedrock formation in much of the upland area of the county. Above and unconformably overlying the Carlile Shale is the Fort Hays Limestone Member of the Niobrara Chalk (Hattin, 1962, p. 91), which is exposed in the extreme northwest corner of the county.







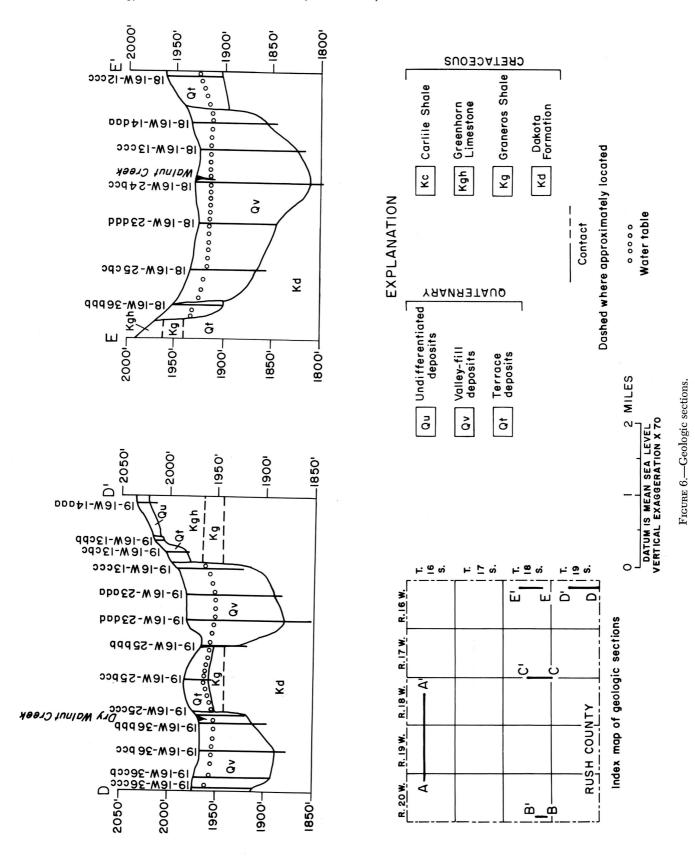


Table 2.—Generalized section of geologic units and their water-bearing characteristics.

System	Series	Group	Formation or stratigraphic unit	Member	Thickness (feet)	Character	Water Supply ¹
			Undifferentiated deposits		0-56	Eolian, fluvial, and colluvial deposits; mostly silt that may overlie sand and some gravel.	May yield small to moderate quantities of water to wells.
Quaternary	Pleistocene		Valley-fill deposits		0-120	Fluvial and eolian deposits of gravel, sand, silt, and clay.	May yield moderate to large quantities of water to wells in Walnut Creek valley and small to moderate quantities of water to wells in other valleys.
			Terrace deposits		0-70	Fluvial and eolian deposits comprised of gravel, sand, silt, and clay.	May yield small to moderate quantities of water to wells.
				Kimball		Fluvial deposits of gravel, sand, silt, and	Occurs on divide areas and
Tertiary	Pliocene		Ogallala Formation	Ash Hollow	0-44	occurrences or son ca algal limestone") and	is not known to yield water to wells.
				Valentine		mg green conglomerate with an opaline matrix.	
_			Niobrara Chalk	Fort Hays Limestone	0-10	Marine deposits of massive chalky limestone.	Yields no water to wells.
				Codell Sandstone		Marine deposits of chalky limestone and	Not known to yield water to
			Carlile Shale	Blue Hill Shale	0-300	careous shale containing large concretions,	wells.
				Fairport Chalk		overlain by friable sifty sandstone. Contains numerous bentonite beds.	
	1	<u></u>		Pfeifer Shale		1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	
	Cretaceous	Colorado	To the section of the section of	Jetmore Chalk	0-100	Marine deposits of alternating chalky limestone and chalky shales; contains bentonite	Not known to yield water to wells.
			Greennorn Limestone	Hartland Shale		Deas.	
_ -				Lincoln Limestone			
Cretaceous			Graneros Shale		0-40	Marine deposits of gray to black noncalcareous to slightly calcareous silty shale with sandstone lenses and bentonite beds.	Not known to yield water to wells.
	1 1 1 1		Dakota Formation	Janssen Clay	-900-300	Some marine but primarily non-marine de-	Yields small to moderate
			Danota i Olimation	Terra Cotta Clay	000	posus or massive wince to prown cray, sur, and shale. Contains interbedded lenticular sandstone and lignite.	quantities of water to wells.
	Lower Cretaceous		Kiowa Formation		100-125	Marine deposits of thinly laminated gray to black shale with sandstone lenses and thin limestone beds in lower part.	Not known to yield water to wells.
			Cheyenne Sandstone		25-100	Some marine but primarily non-marine deposits of buff to gray sandstone with some shale.	Not known to yield water to wells.

Not known to yield water to wells.						
Marine deposits of red siltstone and sand- Not known to yield water to stone with some shale and gypsum. wells.						20-30 Marine deposit of principally anhydrite.
600-700						20-30
Dog Creek Formation Blaine Formation Flower-pot Shale Cedar Hills Sandstone Salt Plain Formation Harper Sandstone						Summer Stone Corral Formation
Nippe- walla						Sumner
Lower						
Lo Permian Pe						

this report, small quantities refers to yields generally less than 10 gpm, moderate quantities to 10 to 500 gpm, and large quantities to greater than 500 gpm

Fluvial deposits of the Ogallala Formation of Pliocene age occur as erosional remnants unconformably overlying Greenhorn Limestone, Carlile Shale, and Fort Hays Limestone Member on the highest upland areas. Clay, silt, sand, and gravel of Pleistocene age unconformably overlie Cretaceous and Pliocene rocks in most of the county. Fluvial deposits are present on divide areas and in stream valleys. Most upland surfaces are thinly mantled with eolian silt or loess. Slope deposits or fluvial and colluvial deposits are present in headland areas of streams and on slopes along stream courses.

GEOLOGY AND WATER SUPPLY OF ROCK UNITS

Rocks that are significant to the water supply in Rush County range in age from Permian to Pleistocene. A generalized section of the rocks and their water-bearing characteristics is given in table 2.

The subsurface Permian and Cretaceous rocks are discussed because of their potential future utilization as aquifers or as confining beds below which brine and other liquid wastes might be disposed. When desalinization processes become more economical and more water is needed in central Kansas, the deeper aquifers could be utilized.

Permian System—Lower Permian Series

SUMNER GROUP

STONE CORRAL FORMATION

The Stone Corral Formation was named from exposures in and near sec. 11, T.20 S., R.6 W., Rice County, Kans., by Norton (1939, p. 1775). Many geologists refer to the Stone Corral as the "Cimarron anhydrite." The Stone Corral is the upper formation of the Sumner Group.

Lithology, distribution, and thickness.—At the type section in Rice County, the Stone Corral is a massive 6-foot ledge of cellular dolomite with some dolomitic shale that forms a prominent scarp (Norton, 1939, p. 1777). The amount of shale increases southward to such an extent that the formation is hardly recognizable in Kingman County (Merriam, 1963). The formation is represented by scattered dolomite rhombs in silty shale in Harper County (Swineford, 1955). The Stone Corral is present in the subsurface westward from Harper County on the south to Jewell County on the north. It is easily discernable with electric logs and is an important marker bed in petroleum exploration. The thickness of the formation in the subsurface is generally between 25 and 45 feet, but in

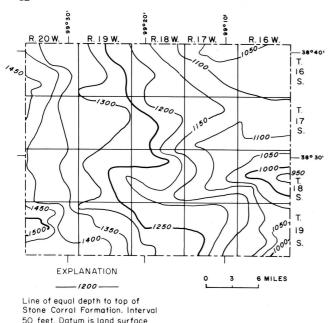


FIGURE 7.—Depth to top of Stone Corral Formation.

Scott County it is 100 feet thick (Norton, 1939, p. 1781). The Stone Corral in Rush County is between 20 and 30 feet thick. Depth to the Stone Corral in Rush County, as determined from electric and radioactive logs, ranges from about 900 to 1,550 feet. Figure 7 is a map showing the approximate depth to the top of the Stone Corral in Rush County.

Water supply.—The Stone Corral is not known to contain water in Rush County; rather, it is regarded as a barrier to the movement of water. Below the Stone Corral, brine and other wastes might be disposed with relatively little chance of leakage upward into higher formations.

NIPPEWALLA GROUP

The Nippewalla Group, which was named by Norton (1939, p. 1782), included several hundred feet of red beds between the Stone Corral and Blaine Formations. The State Geological Survey of Kansas (Zeller, 1968) now includes the following formations, in ascending order, in the Nippewalla Group: Harper Sandstone, Salt Plain Formation, Cedar Hills Sandstone, Flower-pot Shale, Blaine Formation, and Dog Creek Formation.

Lithology, distribution, and thickness.—The Nippewalla Group consists of siltstone and very fine grained sandstone, with minor quantities of silty shale and gypsum (Swineford, 1955). The group crops out in central and south-central Kansas and is in the subsurface in the western half of the State. The thickness of the Nippewalla in Kansas is reported to be approxi-

mately 930 feet by Moore and others (1951, p. 38), from 610 to 870 feet with an original thickness possibly greater than 1,000 feet by Lee (1953, p. 5), and through addition of formation thicknesses from 869 to 948 feet by Swineford (1955). In Rush County the thickness of the Nippewalla probably ranges from 600 to 700 feet.

A study of numerous electric and radioactive logs and some sample logs from Rush County and surrounding counties indicates that the Nippewalla is difficult to subdivide into the accepted stratigraphic sequence. The Harper Sandstone and Salt Plain Formation were not differentiated in this study. What is tentatively called Cedar Hills Sandstone in this report coincides with a subdivision picked as Cedar Hills by the Kansas Sample Log Service and described lithologically as "amber and orange-pink, rounded, polished, friable, fine to coarse, free-drilling sandstone, and shaly sandstone." Lee (1953) described the Cedar Hills in the subsurface as ". . . red sand and siltstone interstratified with silty shale. The sandstones, which are medium- to fine-grained, are in the main orange colored and include many polished subrounded grains slightly coarser than the more angular particles in which they are embedded. Because the sandstone crumbles easily, coherent fragments are rare in the cuttings." Figure 8 is a map showing the depth to the top of the Cedar Hills Sandstone in the county.

The Flower-pot Shale, Blaine Formation, and Dog Creek Formation, which are difficult to identify in the subsurface, were not differentiated in this study. The

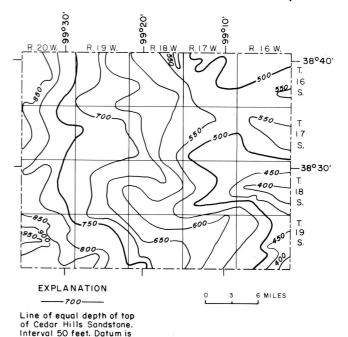


FIGURE 8.—Depth to top of Cedar Hills Sandstone.

part of the Nippewalla Group above the Cedar Hills Sandstone is about 100 feet thick, but may be considably thicker locally; it is relatively impermeable and retards movement of highly mineralized water from the Cedar Hills Sandstone to the overlying Cretaceous rocks.

Water supply.—Sandstone beds in the Nippewalla probably contain water. No data were obtained on the quantity or quality of water available from the Nippewalla in the county. Water from Permian rocks, probably the Nippewalla, was collected from six test holes in Russell County by Swineford and Williams (1945). The concentration of dissolved solids in the samples ranged from 31,000 to 71,000 mg/1 (milligrams per liter).

Cretaceous System—Lower Cretaceous Series

CHEYENNE SANDSTONE

The Cheyenne Sandstone was named by Cragin (1889, p. 65) from Cheyenne Rock in sec. 8, T.30 S., R.16 W., which is a prominent ledge of sandstone on the north side of Medicine Lodge River valley about three-fourths of a mile west of Belvidere in southeast Kiowa County.

Lithology, distribution, and thickness.—The Cheyenne at the outcrop comprises chiefly light-colored friable cross-bedded fine- to medium-grained sandstone associated with lenses of sandy shale and conglomerate (Latta, 1946, p. 235). Swineford and Williams (1945) described the Cheyenne in the subsurface in Russell County as predominantly buff to light-gray sandstone containing small amounts of shale and siltstone. In the subsurface of Cheyenne County, Merriam and others (1959) describle the Chevenne Sandstone as light-gray or brown fine- to medium-grained sandstone with some medium- to dark-gray noncalcareous shale. Exposures of Cheyenne Sandstone are confined to Barber, Kiowa, and Comanche Counties in south-central Kansas (Latta, 1946, p. 238), but the unit may be found in the subsurface in most of western Kansas.

The erosional surface on which the Cheyenne was deposited caused considerable range in thickness of the formation. In the type area the thickness of the Cheyenne ranges from 32.5 to 94 feet (Latta, 1946, p. 238), and in the subsurface the maximum thickness is 300 feet (Moore and others, 1951). In Rush County the thickness is estimated to range from 25 to 100 feet.

Water supply.—The Cheyenne is considered to be an aquifer, but is not known to have been tapped for water supplies in Rush County. To the northeast in Russell County, Swineford and Williams (1945) collected six samples of water from the Cheyenne that contained from 33,000 to 62,000 mg/l dissolved solids.

KIOWA FORMATION

The Kiowa Formation, which overlies the Cheyenne, was named by Cragin (1894, p. 49) from exposure in Kiowa County, Kans.

Lithology, distribution, and thickness.—The Kiowa Formation in the type area was described by Latta (1946) as consisting primarily of dark-gray to black thinly laminated shale grading upward into gray, tan, and brown clay and clay shale, but containing sandstone lenses throughout and thin beds of shell limestone in the lower part. In the subsurface in Russell County, the Kiowa consists of gray to black thinly laminated shale and interbedded thin white sandstone and siltstone (Swineford and Williams, 1945). In the subsurface in Cheyenne County the formation consists of medium- to dark-gray shale, some gray sandstone and siltstone, and a thin bed of impure limestone (Merriam and others, 1959). The Kiowa is exposed over a wide area in central Kansas (Plummer and Romary, 1942) and is in the subsurface in the western half of the state. Latta (1946) reports a maximum thickness of 293 feet of Kiowa in a test hole in the type area. Plummer and Romary (1942) report 100 to 125 feet over much of central Kansas; the range in thickness is assumed to be the same for Rush County.

Water supply.—The Kiowa is considered to be an aquifer but no wells are known to produce water from the formation in Rush County. Eight water samples from the Kiowa in Russell County contained from 33,000 to 59,000 mg/l dissolved solids (Swineford and Williams, 1945).

Cretaceous System— Lower and Upper(?) Cretaceous Series DAKOTA FORMATION

Meek and Hayden (1862) applied the name "Dakota Group" to the sandstone, various colored clay, and lignite beds that underline the "Benton Group" in eastern Nebraska. The name "Dakota Formation," as applied by the State Geological Survey of Kansas, is used in the sense described by Plummer and Romary (1942) and includes the continental and littoral beds occurring above the Kiowa Formation and below the Graneros Shale. The Dakota was subdivided by Plummer and Romary (1942) into the Terra Cotta Clay Member at the bottom and the Janssen Clay Member at the top.

The State Geological Survey of Kansas classifies the Dakota Formation in the Lower and Upper(?) Cretaceous Series; O'Connor (1968, p. 56) states that the boundary between Lower and Upper Cretaceous rocks "properly should be placed within the Dakota Formation".

Lithology, distribution, and thickness.—The following description of the lithology of the Dakota is taken from the description by Plummer and Romary (1947). The Terra Cotta Clay Member, which includes massive clay, silt, and sandstone, composes approximately the lower two-thirds of Dakota Formation. The Janssen Member includes beds of lignite, gray to darkgray massive clay, silt, and some shale. Prominent sandstone layers and lenses are numerous but about three-fourths of the formation is clay, which is quantitatively the most important type of sedimentary material. The clay is dominantly kaolinite.

The Dakota crops out in Kansas from Washington County on the north, southwest to Kiowa County on the south, and in small patches in the southwest part of the State. Except in extreme south and southwest Kansas, it is found in the subsurface west of the outcrop. The Dakota does not crop out in Rush County. In the deep valleys of Walnut Creek and its tributaries in the east half of the county, and the Smoky Hill River and its tributaries in the northeast part of the county, the bedrock underlying the valley-fill deposits is Dakota. Study of electric and radioactive logs did not permit adequate delineation of the top or bottom of the Dakota in Rush County, and test-hole information is sparse. The thickness of Dakota ranges from 100 to 300 feet (Moore and others, 1951), although Merriam and others (1959) report 350 feet in the subsurface in Cheyenne County. In Rush County the thickness is estimated to range from 200 to 300 feet.

Water supply.—The Dakota is a principal aquifer, and dependable water supplies generally are available to wells throughout Rush County. Well yields range from a few gallons per minute for domestic supplies to a reported 500 gpm (gallons per minute) from a municipal well at La Crosse. Water from the Dakota is used extensively in the county except in Walnut Creek and Smoky Hill River valleys and their major tributaries. Reports from well drillers and land owners indicate that the depth to sandstone lenses in the Dakota is unpredictable. The quality of water in the Dakota is variable and generally is considered to be marginal for domestic use and irrigation. The dissolved-solids concentration ranged from 309 to 3,660 mg/l for 30 samples analyzed.

Cretaceous System—Upper Cretaceous Series GRANEROS SHALE

The Graneros Shale was named by Gilbert (1896) from exposures at Graneros and on Graneros Creek in

Pueblo County, Colo. The formation overlies the Dakota and is unconformably overlain by the Greenhorn Limestone in central Kansas (Hattin, 1965).

Lithology, distribution, and thickness.—The Graneros is composed primarily of gray to black noncalcareous to slightly calcareous silty shale; it contains some sandstone lenses and bentonite. One bentonite layer is an easily identifiable marker bed on electric logs in the western third of Kansas (Merriam, 1957).

Surface exposures of Graneros generally are poor, but it is exposed from Ford County on the southwest to Washington County on the northeast. The formation is in the subsurface west of the outcrop belt. No outcrop of Graneros was located in Rush County. The Graneros is present in the subsurface throughout the county, except beneath Walnut Creek and deep tributary valleys east of R.19 W.

Reported thicknesses of the Graneros range from 14 feet in Russell County (Swineford and Williams, 1945) to about 100 feet in Cheyenne County (Merriam and others, 1959). Hattin (1965) reports a range in thickness of the Graneros from 23.6 to 40.4 feet in central Kansas; this range in thickness is assumed to apply to Rush County where the Graneros is present.

Water supply.—The Graneros is not known to yield water to wells in Rush County.

GREENHORN LIMESTONE

The Greenhorn Limestone was named by Gilbert (1896) from exposures at Greenhorn Station and on Greenhorn Creek, south of Pueblo, Colo. Four members of the Greenhorn are recognized by the State Geological Survey of Kansas. In ascending order they are: Lincoln Limestone Member, Hartland Shale Member, Jetmore Chalk Member, and Pfeifer Shale Member. Rubey and Bass (1925) named the Lincoln and Jetmore, and Bass (1926) named the Hartland and Pfeifer. The Greenhorn lies unconformably on Graneros Shale and conformably underlies Carlile Shale.

Lithology, distribution, and thickness.—The Greenhorn comprises a sequence of alternating chalky limestone and chalky shale; it contains thin crystalline limestone beds in the basal member. The outcrop of the formation extends from Ford County on the south to Washington County on the north, with less extensive exposures in Hamilton and Kearny Counties in the southwest. In the subsurface the Greenhorn is found west and north of the outcrop and into adjacent states. Thickness of the formation ranges from 65 feet in Jewell County to 132 feet in Hamilton County. The thickness averages 100 feet in Rush County.

The Greenhorn Limestone is the oldest and most

prominent rock unit that crops out in Rush County. The Greenhorn is not subdivided on plate 1.

Exposures of the Lincoln Limestone Member were not noted during field work, but some may be present in the northeastern part of the county. The Hartland Shale Member is exposed in stream banks, particularly in the northeastern part of the county.

The Jetmore Chalk Member is the most resistant member of the Greenhorn and the "shell" or "shell-rock" bed at the top is diagnostic. The "shellrock" averages about 1 foot in thickness, may be unevenly bedded and appear to have a concretionary upper surface, and is composed chiefly of fossil clam shells. The top bed of the Jetmore has been quarried in the county.

The top bed of the Pfiefer Shale Member, the Fence-post limestone bed, has been extensively quarried. Other limestone beds in the Greenhorn and also in the lower part of the Carlile Shale occasionally have been quarried; thus, reliance on quarries alone to map the top of the Greenhorn is not recommended. Perhaps the best marker for mapping the top of the Greenhorn is the "sugar sand" (Moss, 1932). The "sugar sand" is about 6 feet below the top of the Fence-post Limestone, and about 15 feet above the "shellrock" or top of the Jetmore. White calcite prisms are found at the base of the "sugar sand," which is composed primarily of granular calcite with a silty clay binder. It is crumbly, averages 4 inches in thickness, and resembles white sugar when fresh and brown sugar when weathered. A similar sand about 2 inches thick associated with 2 inches of bentonite, which underlies a 6-inch limestone in the basal Carlile Shale, occurs about 5 feet above the top of the Fence-post bed. The occurrence of these insignificant-appearing sands are quite useful for locating the Greenhorn-Carlile contact.

Water supply.—The Greenhorn Limestone is not considered to be an aquifer in Rush County and no wells were inventoried that obtained water from the formation. Several dug wells in the county terminate in the Greenhorn but the water in the wells comes from overlying colluvial and fluvial material.

CARLILE SHALE

The Carlile Shale was named by Gilbert (1896) from exposures of gray shale near Carlile Station west of Pueblo, Colo., along the Arkansas River. Three members of the Carlile are recognized by the State Geological Survey of Kansas. In ascending order they are: the Fairport Chalk Member (Rubey and Bass, 1925), Blue Hill Shale Member (Logan, 1897), and Codell Sandstone Member (Bass, 1926).

Lithology, distribution, and thickness.—The lower

25 feet of Fairport is similar to the underlying Greenhorn and contains primarily chalky limestone beds and chalky limestone concretion zones intercalated with chalky shale. The rest of the Fairport contains chalky shale and some chalky limestone. Bentonite beds in the Fairport are usable markers and are persistent over much of the outcrop area (Hattin, 1962). The Fairport is distributed on the surface from Washington County in the northeast to Finney and Hamilton Counties in the southwest, and in the subsurface west of the outcrop. Thickness of the Fairport ranges from 90 feet along the Saline River in north Ellis County to 120 feet in Finney and Hodgeman Counties (Hattin, 1962).

Outcrops of the Fairport in Rush County, with a few exceptions, are confined to the lower 25 to 30 feet of the member. The exceptions are middle to upper Fairport, which is exposed along Big Timber Creek and along some of the other tributaries of the Smoky Hill River in northwestern Rush County.

The Blue Hill Shale Member is a dark-bluish-gray slightly silty, clayey shale. The member contains numerous large concretions. Hattin (1962, p. 64) states: "Three kinds occur: Calcareous septarian concretions (most abundant), non-calcareous clay-ironstone concretions, and sandstone concretions (least common)." Many of the concretions are found in traceable, widespread zones, but great numbers have no zonal distribution. The distribution of the Blue Hill in Kansas is similar to the Fairport, but offset to the west. Thickness of the member ranges from 72 feet in Hamilton County to 215 feet in Jewell County.

The Blue Hill probably is absent east of R.19 W. in Rush County because of erosion. It crops out in the northwest and southwest corners of the county. Examples of concretions or their remnants may be found in the SE¼ sec. 19, T.16 S., R.20 W., and the SW¼ sec. 20, T.16 S., R.20 W.

The Codell Sandstone Member is chiefly fine to very fine grained silty sandstone. It is distributed over the State in a manner similar to the rest of the Carlile, but farther west than the lower members. Thickness of the member ranges from half a foot in Jewell County to 31 feet in Ellis County (Hattin, 1962).

The Codell crops out in the SW¼ sec. 6, T.16 S., R.20 W., in Rush County and comprises 4 feet of rusty-orange friable silty sand; it overlies 2 feet of the Blue Hill Shale Member, and is overlain by about 5 feet of the Fort Hays Limestone Member of the Niobrara Chalk. No other Codell was noted in the county.

In Rush County, the maximum thickness of the Carlile Shale is estimated to be 300 feet.

Water supply.—No water supplies are known to be obtained from the Carlile in Rush County. In some

circumstances the Carlile may serve as the container in a dug well but the water comes from overlying colluvial and fluvial material. The Carlile is not considered to be an aquifer in Rush County.

NIOBRARA CHALK

The Niobrara Chalk was named by Meek and Hayden (1862) from exposures in bluffs along the Missouri River in northern Nebraska. Two members of the Niobrara are recognized by the State Geological Survey of Kansas—the Smoky Hill Chalk Member above and the Fort Hays Limestone Member below. The Smoky Hill is not present in Rush County and only about 10 feet of the basal Fort Hays Member is present in the northwest corner of the county. The Niobrara Chalk is not an aquifer in Rush County.

Tertiary System—Pliocene Series

OGALLALA FORMATION

The Ogallala Formation was named by Darton (1899) from exposures that he later (Darton, 1920) designated as being near Ogallala Station in western Nebraska. Three members of the Ogallala are recognized by the State Geological Survey of Kansas: the Valentine Member at the base, the Ash Hollow Member in the middle, and the Kimball Member at the top (Zeller, 1968). A detailed discussion of the Ogallala is contained in Bulletin 118 of the State Geological Survey of Kansas (Frye, Leonard, and Swineford, 1956). The Ogallala was mapped as a single unit for this report, but in the following discussion the members are designated when applicable.

Lithology, distribution, and thickness.—The dominant types of lithology of the Ogallala in Rush County are (1) arkosic sand and gravel with associated clay balls that are sometimes poorly cemented with calcium carbonate, clay and silt beds, and some bentonite, (2) a quartzitic-appearing green conglomerate with an opaline cement matrix, and (3) a distinctive, discontinuous, very hard and dense white to pink limestone commonly referred to as "algal limestone." A study by Swineford, Leonard, and Frye (1958) indicates that development of the limestone was by soil-forming processes rather than by algae, and that pisolitic limestone is an accurate descriptive term for the bed. Common conversational usage is "algal limestone," which is used in this report. The "algal limestone" marks the stratigraphic top of the Ogallala Formation and the Kimball Member.

The Ogallala lies unconformably on Greenhorn, Carlile, and Niobrara in Rush County. Its areal distribution, shown on plate 1, generally is confined to the divide areas between Walnut Creek and the Smoky Hill River drainage in the north, and between Walnut Creek and the Pawnee River drainage in the south.

Frye, Leonard, and Swineford (1956, p. 73) describe the following measured section in the SW¼ sec. 1, T.16 S., R.21 W., Ness County:

Th	ickness, feet
OGALLALA FORMATION	1000
Ash Hollow Member	
6. Silt, densely cemented, pale-gray to pinkish-	
gray; contains irregular opalized nodules	3
5. Sand and silt, loosely cemented, pinkish-tan	
partly covered. At this stratigraphic position	1
in the NW_4SW_4 sec. 21, T.15 S., R.21 W.	
Trego County, the following fossil seeds were)
collected: Berrichloa conica, B. tuberculata	
Biorbia fossilia, Celtis willistoni	2.5
4. Sand, fine to coarse, some pebbles, cemented weathers to smooth convex surfaces	7.5
3. Covered	. 7.5 7
Valentine Member(?)	
2. Conglomerate, densely cemented with opal;	
the opaline matrix has distinct green color and	
and is lenticular in distribution	10
1. Gravel and cobbles of Cretaceous Niobrara	
Chalk to 1 foot in diameter, in matrix of clay	
and chalk sand. Base rests on Niobrara Chalk	3
Total Ogallala measured	33

The Ogallala exposed in the northwestern part of Rush County (SW¼ sec. 6, and NW¼ sec. 7, T.16 S., R.20 S.) resembles bed 2, and is locally referred to as "green granite." Except for bed 2, the Ogallala north of Walnut Creek probably represents the Ash Hollow Member. About 20 feet of crossbedded, slightly consolidated arkosic sand and gravel, which rests on upper beds of the Fairport Chalk Member of the Carlile Shale in a gravel pit in the SE¼ sec. 16, T.17 S., R.19 W., is the most extensive exposure of Ash Hollow north of Walnut Creek. Test hole 17-19W-21aaa in the northeast corner of section 21 in the same township penetrated 28.0 feet of Ogallala. Another exposure of Ogallala in the north half of the county is in a trench silo in the NW¼ sec. 12, T.17 S., R.18 W., where about 6 to 8 feet of sand, gravel, and silt rests on the Fairport.

Test-hole logs and samples, shot-hole logs, gravel pits, and natural exposures document the occurrence of considerable Ogallala, predominantly Ash Hollow but with some Kimball, south of Walnut Creek. Gravel pits in the SW¼ sec. 8, T.19 S., R.20 W.; the SE¼ sec. 28, T.19 S., R.19 W.; and the SE¼ sec. 26 and the SW¼ sec. 25, T.19 S., R.18 W., contain slightly consolidated very fine to medium-grained crossbedded sand and some gravel of arkosic composition with associated balls of clay, and some local gravel composed of Cretaceous limestone fragments. The thickest Ogallala found in the county was about 44 feet in test hole 19-18W-33aba. The Kimball is recognized in numer-

ous places south of Walnut Creek on the basis of "algal limestone" that caps topographic highs, including distinctive conical knolls. "Algal limestone" is exposed in the trail between the SE¼ sec. 36, T.18 S., R.16 W., and the NE¼ sec. 1, T.19 S., R.16 W.; in the south road ditch in the NE¼NW¼ sec. 16, T.19 S., R.16 W.; on top of the hill in the NW¼ sec. 25, T.19 S., R.18 W.; on top of the hill in the SE¼ sec. 28, T.19 S., R.18 W.; toward the top of two trench silos on the north side of the road in the SE¼SE¼SW¼ sec. 29, T.19 S., R.19 W.; on a small ridge south of the road in the NW¼ sec. 17, T.19 S., R.20 W.; along the road in the NW¼ sec. 26, T.19 S., R.20 W.; and in numerous other places in the south tier of townships. The lithology and thickness of the Ogallala are shown by logs at the end of this report.

Water supply.—The Ogallala is not an aquifer in Rush County because it is drained by streams that have eroded headward toward the divide areas. The drainage precludes water storage except in the most temporary sense. None of the wells inventoried obtained water from the Ogallala.

Quaternary System—Pleistocene Series

The Pleistocene Series has been divided into four glacial stages (Nebraskan, Kansan, Illinoisan, Wisconsinan) and four interglacial stages (Aftonian, Yarmouthian, Sanagmonian, Recent) by the State Geological Survey of Kansas (Bayne and O'Connor, 1968). Most of these stages are thought to be represented in Rush County. Topographic position, fauna, and lithology of deposits in relation to adjoining or very similar deposits in adjacent counties form the basis for this belief. Three categories of Pleistocene deposits were mapped in the county (pl. 1). Two of the three mapped units are unconsolidated, principally fluvial, deposits conveniently separated by their relative topographic position. The deposits in terrace position, herein called terrace deposits, and the deposits in valleys, herein called valley-fill deposits, may be of Kansan, Illinoisan, Wisconsinan, or Recent age. The third mapped unit is primarily an unconsolidated eolian deposit but contains some fluvial and colluvial material. These deposits, herein called undifferentiated deposits, may be as old as Nebraskan but commonly are Wisconsinan and Recent in age.

TERRACE DEPOSITS

Lithology, distribution, and thickness.—Fluvial deposits with some eolian and colluvial components are found in terrace position along the valleys of the major streams and their tributaries. These deposits of clay, silt, sand, and gravel primarily are arkosic, but often

include some limestone pebbles. Thickness of the deposits may be as much as 70 feet bordering Walnut Creek valley, 50 feet bordering Big Timber Creek valley, and 40 feet bordering Dry Walnut Creek valley.

Water supply.—The terrace deposits yield small to moderate supplies of water to wells. Two irrigation wells in Walnut Creek valley and one in Dry Walnut Creek valley obtain all or part of their water from these deposits; yields range from 290 to a reported 400 gpm. Numerous stock and domestic wells also obtain their water from these deposits. Water from the terrace deposits is hard, but otherwise is of satisfactory chemical quality for the uses mentioned.

VALLEY-FILL DEPOSITS

Lithology, distribution, and thickness.—Fluvial clay, silt, sand, and gravel compose the valley-fill deposits. The upper 20 to 40 feet of the fill is predominantly silt with some clay that overlies a thick deposit of sand and gravel in Walnut Creek valley. In Dry Walnut Creek valley, the fill is principally silt with minor amounts of sand and gravel. The fill in Big Timber Creek valley is not as thick as the fill in Walnut Creek valley, but the lithology is similar. Thickness of the fill may be as much as 120 feet in Walnut Creek valley, slightly more than 100 feet in Dry Walnut Creek valley, and more than 40 feet in Big Timber Creek valley. The valley fill is as much as 2 miles wide in Walnut Creek valley and three-fourths of a mile wide in Dry Walnut and Big Timber Creek valleys.

About 30 different forms of snails and several forms of clams were noted in the samples from test holes drilled across Walnut Creek valley. Samples from test hole 18-17W-31bbc were particularly abundant in number of shells and in number of forms. The precise depths from which the different shells came are not known, but, between 30 and 75 feet, shells were so numerous as to appear to be the major part of the return from the drilling operation.

The drill cuttings containing the shells were blue to gray silty clay, fine to coarse arkosic sand, and medium to coarse, predominantly limestone, gravel. Although the volume of material returned in the 30-to 75-foot interval was normal, only 10 small sacks of sample material were collected. Each sack of the material was then split into one small sample-cuttings envelope in the office. In light of the sampling procedure just described, the number of shells and forms recovered is remarkable.

The following fossil mollusks were recovered from the sample cuttings from test hole 18-17W-31bbc, Rush County, Kans. The shells were identified by the author.

SPECIMENS Fresh-water snails Amnicola limosa parva Lea 44 Ferissia parallela (Haldeman) 3 Gyraulus labiatus Leonard 112 Helisoma antrosa (Conrad) 7 Lymnaea SP Menetus pearlettei Leonard Pomatiopsis cincinnatiensis (Lea) 2 Valvata tricarinata (Say) 177 LAND SNATES Carychium perexiguum Baker _____ 6 Gastrocopta proarmifera Leonard 5 Gastrocopta procera (Gould) 1 Gastrocopta tappaniana (C.B. Adams) 2 Helicodiscus parallelus (Say) 2 Pupilla muscorum sinistra Franzen Pupoides albilabris (C. B. Adams) 2 Retinella electrina (Gould) 12 Succinea SP 17 Vallonia gracilicosta Reinhardt 72 Vertigo millium (Gould) 4 Vertigo ovata Say 2 Zonitoides arboreus (Say) 3 AT LEAST TWO FORMS OF CLAMS

The fauna indicates that the summers were not as hot as they are in Rush County at the present time, but winters may not have been cooler. Precipitation probably was greater and a generally more humid condition prevailed. Permanent bodies of cool water are indicated. In terms of age, the list above appears to fit well in the assemblage of fossils of Kansan age described by Leonard (1950) and Frye and Leonard (1952).

Water supply.—Yields ranging from 400 to 1,200 gpm are obtained from the valley-fill deposits by irrigation wells in Walnut Creek valley, and yields as large as 1,500 gpm have been reported. No irrigation wells obtain water from these deposits in Big Timber and Dry Walnut Creek valleys; however, small to moderate supplies are obtained by stock and domestic wells. The water is hard but suitable for most uses. The dissolved-solids concentration for 17 water samples analyzed ranged from 369 to 789 mg/1.

UNDIFFERENTIATED DEPOSITS

Lithology, distribution, and thickness.—Eolian deposits, with some fluvial and colluvial components that are predominantly silt containing some sand and some gravel, compose the undifferentiated deposits. The silt probably is loess of Wisconsinan age; it commonly overlies sand and some gravel. In the northwestern part of the county, the lower part of the undifferentiated deposits may be of Nebraskan age because of their topographic position and relation to similar de-

posits in Ellis County. The greatest thickness of these deposits, as much as 56 feet, was found in Tps. 17 and 18 S., R.20 W. Other areas with thick undifferentiated deposits are in T.16 S., R.20 W., and T.17 S., R.16 W. These thick deposits definitely are fluvial in origin, except for the overlying loess, and may represent buried channels of Nebraskan age.

Water supply.—Small to moderate yields are obtained from these deposits by some stock and domestic wells in the county where the deposits are relatively thick. However, such a favorable circumstance is not widespread. The water is very hard, but otherwise is satisfactory for stock and domestic uses.

GROUND WATER

The following discussion of ground water in Rush County is devoted principally to the aquifer in Walnut Creek valley, which comprises the valley-fill and terrace deposits. Definitions of the terms used in the discussion are given in the Glossary of Terms.

Source

NUMBER OF

The primary source of ground water in Rush County is local precipitation. For example, the ground water in the valley-fill deposits of Walnut Valley is derived from the rain and snow on the entire watershed of Walnut Creek. Water in the Dakota Formation, on the other hand, is derived from rain and snow on the outcrop of the formation east of Rush County, and from unconsolidated water-saturated deposits, such as those in Walnut and Dry Walnut Creek valleys, that overlie the Dakota. Ground water, thus, consists of water that falls in the form of rain or snow and then percolates through the soil materials to the water table. Figure 9 illustrates the relationship between precipitation, ground-water levels in selected wells, and streamflow in Rush County.

Occurrence and Movement

Water in the unconsolidated deposits occurs in the interstices between the rock particles. The rate at which water will move through these aquifers depends on the hydraulic gradient and on the shape, size, and interconnection of the contained interstices. The quantity of ground water available to wells also depends on the areal extent and the saturated thickness of the aquifer. Interstices in sand and gravel are larger and better connected than interstices in silt and clay. Thus, water will move freely through a coarse gravel under a low hydraulic gradient, but will move with extreme slowness through clay even under a high hydraulic gradient.

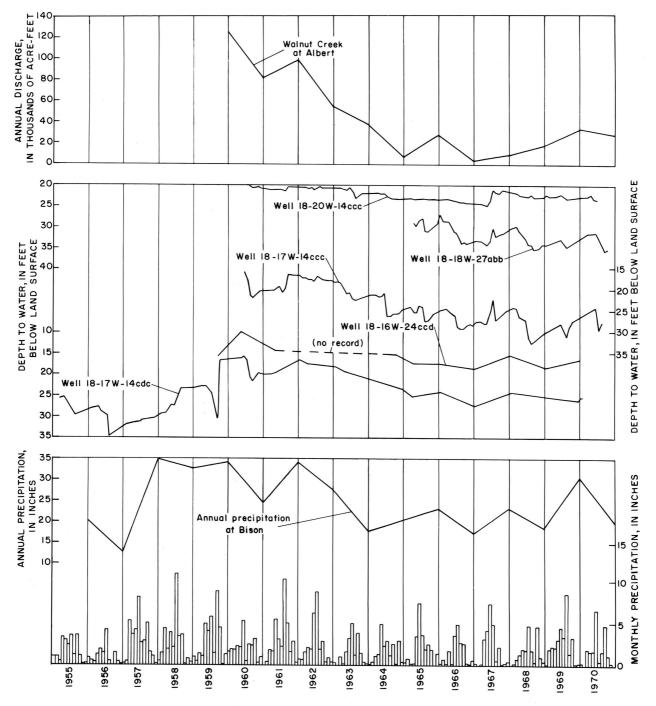


Figure 9.—Change in water level in five wells in Walnut Creek valley, monthly precipitation at Bison, and annual discharge of Walnut Creek at Albert.

The shape and slope of the water table in 1960 in Walnut Creek valley are shown on plate 1 by contours drawn through points of equal altitude of the water table. Ground water moves downgradient at right angles to the contours. The contours indicate that ground water was moving toward Walnut Creek from both sides of the valley and was discharging into the

creek in 1960. The spacing of contours in Walnut Creek valley indicate a hydraulic gradient of about 5 feet per mile at the west boundary of Rush County to 3 feet per mile at the east boundary.

The quantity of water flowing through a given cross-sectional area of an aquifer can be computed by the formula:

Table 3.—Calculated velocity and quantity of water moving through the aquifer at the west and east ends of Walnut Creek valley.

							Quantity o	f water
	Average aquifer saturated thickness (feet)	Assumed porosity (percent)	Average aquifer width (miles)	Hydraulic gradient (feet per mile)	Hydraulic conductivity (feet per day)	Velocity (feet per day)	(Cubic feet per day)	(Acre-feet per year)
West end	20	30	1	5	382	1.2	38,200	320
East end	70	30	2.5	3	221	.4	116,000	970

Q = pAv = KIA

where Q is the quantity of water,

p is the porosity of the aquifer material,

A is the cross-sectional area,

v is the average velocity of ground water,

K is the hydraulic conductivity, and

I is the hydraulic gradient.

The approximate rate of movement of ground water through an aquifer is obtained by transposition of the above formula to:

$$v = KI$$
 p

and for the units of feet, days, and miles to:

$$v = \frac{KI}{53 p}$$

In the case of the aquifer in Walnut Creek valley, the amount of water entering Rush County as subsurface inflow at the Ness County line and the amount leaving as subsurface outflow at the Barton County line can be calculated. Additionally, the velocity of the ground water can be calculated. The results of these calculations are given in table 3.

Storage

The amount of recoverable water stored in an aquifer is a function of the volume of the aquifer and its storage coefficient (S).

Maps showing the configuration of the bedrock surface and the saturated thickness of the valley-fill and terrace deposits of Walnut Creek valley in 1960 are shown on plate 2. On the saturated thickness map the area between each pair of contours was planimetered and multiplied by the average saturated thickness to determine the volume of saturated material. The volume of saturated deposits was calculated to be 1,610,000 acre-feet. Based on an assumed storage coefficient of 0.15, the volume of water available for pumping in 1960 was determined to be 241,500 acre-feet. From a practical point of view, much less would be available for irrigation. Yields from wells would

diminish substantially when less than half this quantity of water was pumped from storage.

Changes in Storage

One method of assessing changes in the amount of ground water in an aquifer involves periodic water-level measurements, construction of water-level-change maps from the measurements, and computation of the volume of material and water involved in the change. Unless outside influences such as heavy pumping upset natural conditions, the changes in storage in an aquifer reflect seasonal changes in precipitation and evapotranspiration. Water-level-change maps would illustrate, by minor fluctuations and trends, the essentially static conditions in an undisturbed aquifer. The aquifer in Walnut Creek valley is heavily pumped, and the change maps reveal the effect of large withdrawals of ground water for irrigation.

Changes in storage in the aquifer in Walnut Creek valley are documented by six water-level-change maps (pl. 2). Water-level measurements made in the spring of 1965 and each January from 1966 through 1970 are compared with the base measurements made in the spring of 1960. The maps were planimetered and the changes in volume of saturated material and volume of water were computed (table 4). The changes are

Table 4.—Changes in volume of saturated material and water in storage in Walnut Creek valley from 1960 to indicated year. [Plus (+) and minus (-) indicate increase or decrease in storage, respectively.]

	Decrease in volume of saturated material (acre-feet)	Decrease in volume of water S = 0.15 (acre-feet)	Change in volume of water between successive years (acre-feet)
1965	257,000	38,550	-
1966	259,000	38,850	_ 300
1967	340,000	51,000	-12,150
200.	310,000	32,000	+16,650
1968	229,000	34,350	17.050
1969	344,000	51,600	17,250
1970	271,000	40,650	+10,950

all declines when related to the 1960 measurements, but when successive changes are compared, there are 3 years of declines and 2 years of gains.

The volume of water in storage during the time from the spring of 1965 through January 1970 ranged from 189,900 to 207,150 acre-feet.

Discharge

Ground water in Walnut Creek valley is discharged from the aquifer by pumping by wells, evapotranspiration, subsurface outflow, and inflow to streams. In the section on movement of water, the subsurface outflow from the valley was estimated to be about 970 acrefeet per year.

PUMPING BY WELLS

Pumpage from irrigation wells is reported to the Kansas State Board of Agriculture by irrigators. A summary of the estimated annual pumpage from the aquifer in Walnut Creek valley for the years 1958-67, based on the amounts reported, is given in table 5. The average annual pumpage during this period was 14,100 acre-feet.

Table 5.—Estimated annual pumpage in Walnut Creek valley, 1958-67.

Year	Pu (ac	mpage re-feet) Y	ear	Pumpage (acre-feet)
1958		5,800 19	63	16,800
1959	1	3,300 19	64	18,900
1960		9,900 19	65	16,000
1961		9,700 19	66	21,700
1962		7,300 19	67	•

EVAPOTRANSPIRATION

Direct evaporation is limited to areas where the water table is near the land surface, such as along stream banks and in streambeds. This condition is restricted to a small part of the valley, and discharge by evaporation probably is not large.

Transpiration by plants from the saturated zone is not confined to the water courses where willow, cottonwood, and other trees are concentrated. Alfalfa may transpire considerable quantities of water from the aquifer. No quantitative estimate of evapotranspiration was made.

INFLOW TO STREAMS

Prior to the development of irrigation, Walnut Creek probably was a gaining stream through Rush County except during times of flooding. Examination of streamflow records at Albert, in Barton County 1 mile

east of Rush County, indicates an alternating sequence of gaining and losing flow that reflects precipitation and pumping by wells. Streamflow at Albert, which is summarized in table 6, does not all come from ground water. The recorded flow represents two components: (1) overland flow directly from precipitation, and (2) inflow from ground water. No quantitative estimate of inflow to streams was made.

Table 6.—Summary of streamflow of Walnut Creek at Albert.¹

		Mea (cubi	Annual		
Year		Maximum	Minimum	Mean	streamflow (acre-feet)
1959		10,300	0.2	174	126,000
1960		3,850	.5	113	81,860
1961		3,450	0	136	98,360
1962		1,770	9.6	75.2	54,480
1963	***************************************	1,520	0	50.3	36,400
1964		251	0	9.24	6,700
1965		1,090	0	36.7	26,540
1966		232	0	3.3	2,390
1967		2,600	0	134	96,890
1968		1,830	0	23.7	17,190
1969		1,690	0	43.9	31,750
1970	=======================================	2,820	0	36.0	26,090

¹ From U.S. Geological Survey (1962, 1969, 1967-72)

Recharge

The aquifer in Walnut Creek valley is recharged by subsurface inflow from the west and tributary valleys on the north and south, and by seepage losses from Walnut Creek and tributary creeks during periods of high flow. These two increments of recharge result from local precipitation within Walnut Creek basin.

SUBSURFACE INFLOW

In the section on movement of water, subsurface inflow to the aquifer in Walnut Creek valley from the west was estimated to be 38,200 cubic feet per day, or 320 acre-feet per year. The combined subsurface inflow from Sandy, Otter, and Dry Creek valleys and Old Maid Fork valley on the south and Sand Creek valley on the north probably is at least an equivalent amount, although no formal quantitative determination of inflow from tributary valleys was made.

SEEPAGE LOSSES FROM STREAMS

The drainage system of Walnut Creek basin carries large amounts of surface water into the valley. For this water to percolate to the saturated zone, the surface of the stream must be above the water table, and the sediments between the stream channel and the water table must be permeable. The sediments in

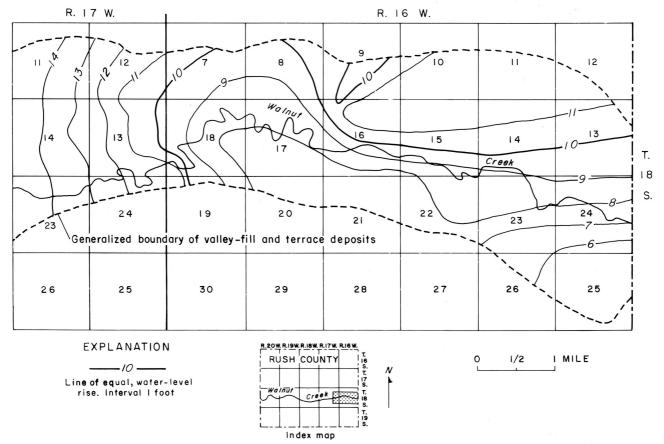


Figure 10.—Water-level rise in valley-fill and terrace deposits in Walnut Creek valley, eastern Rush County, September 1959 to May-June 1960.

Walnut Creek valley are permeable and the water table is below stream level after or during the irrigation season, after or during drought, or during some combination of these conditions. When a major flood occurs, the circumstances for recharge are enhanced because the stream may leave its bank and spread completely across the valley.

A large amount of recharge occurred during the flood of September 1959. This flood, which was the largest of record for Walnut Creek, spread water completely over the valley. The author had measured water levels in 21 wells in the first half of September 1959. The same wells were remeasured in the spring of 1960. A map showing change in water level for part of the valley in eastern Rush County (fig. 10) was constructed and analyzed. The volume of saturated material and the volume of water in storage increased by 104,000 and 15,600 acre-feet, respectively. The surface area of the valley included on the map was about 11,700 acres. Thus, about 1½ feet of water was recharged by the flood of September 1959 to each acre of valley in the mapped area. If this amount of recharge is applied to the entire area of the valley inundated by the flood in Rush County, the total increase in storage would be about 50,000 acre-feet.

The changes in volume of water between successive years (table 4), when combined with the estimated annual pumpage (table 5), also indicate large amounts of recharge to the aquifer (table 7). Table 7 gives the change in volume of water in the aquifer, the estimated volume of water pumped from the aquifer, and the estimated volume of recharge to the aquifer. The volume of recharge is calculated by adding algebraically the volume of change to the volume pumped.

This approach to determination of recharge is dependent on estimated pumpage based on reported

Table 7.—Change in volume of water in, volume of water pumped from, and volume of water recharged to the aquifer in Walnut Creek valley for 1965-69.

Year	Change (acre-feet)	Pumpage (acre-feet)	Recharge (acre-feet)
1965	_ 300	16,000	15,700
1966	-12,150	21,700	9,550
1967	+16,650	21,400	38,050
1968	-17,250	120,000	2,750
1969	+10,950	120,000	30,950
Cumulative			•
Total	- 2,100	99,100	97,000

¹ Assumed values

usage, an estimated storage coefficient, and the assumption that the stream was losing and not gaining during periods of high flow. The principal evidence for a losing stream is the extensive periods of no flow at Albert.

Table 8.—Decrease in volume of saturated material in Walnut Creek valley, 1960-70.

Year	Volume of saturated material (acre-feet)	Percent of 1960 volume of saturated material	Percent de- crease in 1960 volume of sat- urated material	Percent change in volume of saturated material between successive years
1960	1,610,000			
1965	1,353,000	84	16	
1966	1,351,000	84	16	0
1967	1,270,000	79	21	–5
1968	1,381,000	86	14	+7
1969	1,266,000	79	21	7
1970	1,339,000	83	17	+4

Even if the estimates for pumpage and storage coefficient are not considered to be of the right order of magnitude, the changes in volume of saturated material are real. The cumulative change in volume of saturated material for the 5-year period 1965-69 is a decline of only 14,000 acre-feet (2,100 acre-feet of water ÷ 0.15 storage coefficient). Table 8 expresses the change in volume of saturated material in percent of volume of saturated material in 1960. Beginning in 1965 the volume of saturated material has changed by as much as 7 percent of the 1960 total in 1 year. This indicates that significant recharge occurs, and occurs rapidly. The drainage network of Walnut Creek, which funnels runoff into the valley, thus concentrating precipitation, is the most effective recharge mechanism to the aquifer.

HYDROLOGIC PROPERTIES OF WATER-BEAR-ING MATERIALS IN WALNUT CREEK VALLEY

The quantity of water available from an aquifer depends on the ability of the aquifer to store and to transmit water. The ability of an aquifer to store water is measured by its storage coefficient and the ability to transmit water by its transmissivity. These hydraulic properties are in turn dependent upon the dimensional and geological parameters of the aquifer.

Three aquifer tests were made to determine the storage coefficient, the transmissivity, and the hydraulic conductivity of the valley-fill deposits in Walnut Creek valley. The test data were analyzed by methods referred to as the Thiem method, the Theis nonequilibrium method, and the Jacob modified nonequilibrium method (Ferris and others, 1962; Stramel, Lane, and Hodson, 1958). Each test was analyzed by all or some of the preceding methods in an attempt to arrive at aquifer coefficients judged to be most nearly correct. The theory and mathematical derivations of the methods are presented in the references noted.

Results of analysis of the tests are presented in table 9. Under the heading of hydrologic properties both the terms previously used by the U.S. Geological Survey, coefficient of transmissibility and field coefficient of permeability, and the terms currently used, transmissivity and hydraulic conductivity (Lohman and others, 1972), are listed.

The storage coefficients determined by the three aquifer tests range from 1.4×10^{-1} (unconfined aquifer) to 2.6×10^{-4} (confined aquifer). The disparities in storage coefficients could result from partial penetration of the aquifer by the pumping well or the observation wells, localized confining layers in the aquifer, insufficient length of aquifer tests, or some combination of these or other factors. The aquifer is considered to be generally unconfined in Walnut Creek valley and to be similar to aquifers in other stream valleys where storage coefficients of 0.15 and 0.20 have been used.

Table 9.—Aquifer-test results.

			Hydrologic properties					
Well number and owner	Date of test (1960)	Duration of test (minutes)	Coefficient of transmissibility (gal per day per ft)	Transmissivity (ft² per day)	Field coefficient of permeability (gal per day per ft²)	Hydraulic conductivity (ft per day)	Storage coefficient (dimensionless)	Yield (gpm)
18-16W-16bbb (Glantz)	August 3-5	2,733	100,000	13,400	1,650	221.1	$2.6 \times 10^{-4} \ 3.6 \times 10^{-2}$	1,000
18-17W-28bcc (Oborny)	August 15-22	10,095	70,000	9,380	1,750	234.5	$3.1 imes 10^{-2} \ 4.4 imes 10^{-3}$	740
18-19W-21cdb (Webs)	July 16-17	1,840	100,000	13,400	2,850	381.9	1.2×10^{-2} 1.4×10^{-1}	450

Table 10.—Chemical analyses of water from selected wells, Walnut Creek, and one seep¹ [Dissolved constituents and hardness given in milligrams per liter.²]

							,	•
So- dium- ad- sorp- tion ratio	22 23 26 31	31 16 28 .5	223324 233824 833824	18 11 9.0 23 .5	1.7 .6 1.2 13 .6	1.2 21 .5 .5 .8	27 1.5 6.1 .7	11. 1.4 .3 .7.
Нq			8.0	7.8	7.6		7.8	
Specific conduct- ance (micro- mhos at 25 °C)	2,650 2,170 2,680 3,870 29,800	3,430 $1,800$ $2,320$ 700 $1,870$	3,320 2,370 6,740 2,370 2,370	2,400 3,370 1,900 1,640 590	1,270 760 730 1,310 690	790 2,360 590 720 1,160	2,470 930 920 630 700	1,160 $1,050$ 710 740 850
ness as 50 ₃ Non car- bonate	0 0 0 0 6,300	00090	0000	84 26 0 22 22	110 46 36 0 78	36 0 32 57 0	$\begin{smallmatrix}0&4&0&4\\0&0&2&4\\6&0&1&3\end{smallmatrix}$	$\begin{array}{c} 0 \\ 120 \\ 66 \\ 68 \\ 140 \end{array}$
Hardness CaCO ₃ Cal- cium, No Mag-can nesium bons	$\begin{array}{c} 100 \\ 72 \\ 72 \\ 72 \\ 110 \\ 6,500 \end{array}$	92 86 300 24	$\begin{array}{c} 140 \\ 46 \\ 210 \\ 78 \\ 56 \end{array}$	120 440 270 36 270	490 330 280 84 300	310 94 270 300 84	58 350 110 330 290	36 400 320 340 380
Ni- trate ⁵ (NO ₃)	3.4 1.6 2.8	3.8 3.8 18 2.1	4.9 1.9 1.9 1.2	0.4.8.E.R.	7.1 6.2 1.3 5.8	9.4 9.3 1.9	1.8 6.2 4. 4.	1.8 18 34 6.2 23
Fluo- ride (F)	8.8.8.8. 8.4.4.7.	4.00.4 4.00.04	3.3.8 3.2.8 3.7	0.4.9.0. 61.18.6161	r: ci ro ro 4	3.0 3.0 5.2 5.3	2.4. 2.4.5. 2.5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.	e ಜ್ಞಾಪಪ್ರಪ್ರಪ್ರ
Chlo- ride (Cl)	510 440 540 880 $9,600$	700 240 360 39 290	700 500 1,800 430 440	430 580 260 240 19	57 38 31 46 76	24 460 15 22 70	440 46 84 19 27	110 49 17 23 34
Sul- fate (SO ₄)	210 200 190 260 1,400	270 180 220 37 88	240 190 290 220 220	210 450 360 160 30	190 48 80 140 17	93 140 40 85 120	230 99 52 53 47	95 180 40 90 140
Bicar- bonate (HCO ₃)	332 277 274 307 190	356 398 431 290 442	321 273 249 294 284	307 432 298 268 304	458 346 296 554 271	337 336 288 300 407	298 378 329 351 300	354 340 316 334 293
So- dium ⁴	510 460 510 760 4,000	680 360 480 20 390	640 500 1,300 460 480	450 550 340 320 19	84 28 270 24	46 460 18 31 220	480 62 150 19 28	240 66 13 29 32
Mag- nesium (Mg)	13 11 9.6 19 360	11 16 9.6 20 1.3	22 5.1 29 11 5.1	9.4 46 36 3.3 9.1	19 11 15 11	12 10 8.8 10 9.5	$\begin{array}{c} 5.6 \\ 11 \\ 10 \\ 11 \\ 7.3 \end{array}$	2.7 12 9.0 11 10
Calcium (Ca)	$\begin{array}{c} 19\\11\\13\\13\\14\\2,000\end{array}$	19 8.3 8.3 89 7.4	18 10 38 13 14	$\begin{array}{c} 31 \\ 100 \\ 49 \\ 9.1 \\ 94 \end{array}$	160 110 95 95 9.1	100 21 93 100 18	14 120 29 114 100	10 140 120 120 140
Total man- gan- ese (Mn)	0.00 00.00 00.00	000000	00.00.00.00.00.00.00.00.00.00.00.00.00.	<u> </u>	Ŗij	.70 .00 .31 .00	.00 .00 .42	66666
Total iron (Fe)	.21 .32 .56 .02	.42 .07 .01 .54	.16 .09 .30 .17	$\begin{array}{c} 1.2 \\ .08 \\ .10 \\ 2.7 \\ .18 \end{array}$	32 32 36 36 27	1.8 .46 .20 .38	1.1 1.5 .36 .62 .18	1.1 .18 .18 .29
Silica (SiO ₂)	0 10 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	8.0 8.0 8.5 46 6.0	7.0 8.0 8.0 7.5	$\begin{array}{c} 9.0 \\ 6.0 \\ 10 \\ 6.0 \\ 47 \end{array}$	34 31 28 8.0 50	36 9.0 43 41 8.0	8.0 8.0 8.0 35	9.0 33 31 45 36
Dis- solved solids (evapo- rated at 180 °C)	1,440 $1,270$ $1,410$ $2,090$ $17,400$	1,870 1,030 1,300 412 1,000	1,790 1,360 3,660 1,300 1,320	1,300 1,950 1,200 884 374	789 444 441 773 420	487 1,270 369 446 650	1,320 568 500 415 420	644 665 414 488 549
Tem- pera- ture (°C)	16.0 15.5 16.5 15.5 13.0	14.0 18.5 14.5	15.5	16.5 17.0 15.0	$13.0 \\ 22.0 \\ 14.5 \\ 13.0$	12.0 12.0 12.0	12.0 12.0	14.5 12.0 12.0 13.0
Date of col- lection	5-12-60 5-12-60 10- 6-60 5-13-60 5-13-60	5-12-60 5-13-60 5-13-60 5-13-60 10-13-60	5-12-60 10-12-60 5-12-60 4- 1-60 5- 5-59	5-8-61 5-16-60 5-13-60 10-12-60 8-5-60	8-19-60 5-25-61 8-24-60 5-17-60 8-17-60	8-19-60 10- 7-60 8-24-60 8-11-60 10-11-60	10-12-60 8-22-60 10-12-60 7-22-61 8-11-60	10-11-60 8-22-60 8-19-60 8-11-60 8-11-60
Geologic source ³	Kd Kd Kd	Kd Kd Qu Kd	Kd Kd Kd Kd	Kd Kd Kd Qv	Or Kd	o Q Q Q	K Q Q Q Q	M O O O O O O
Depth (feet)	266 370 320 230 seep	209 400 530 315	230 335 300 300	302 320 325 86	55 75 stream 168 86	70 224 59 63 175	265 67 187 83 70	150 539 54 58 64
Sample number (figs. 11-13, 18)	H00470	6 8 9 10	112121	16 17 18 19 20	22222 2223 3	30 30 30 30	33 33 34 35 35	36 337 339 40
S n Nell number	16-16W-32ccb 16-17W-16dcd 22aaa 16-18W- 6dac 6dac2	27cdd 16-19W-17bab 16-20W-5bbd 21ddc 17-16W-20aaa	17-17W-28dcd 17-18W-20daa 28aad 33acd 33acd	34bbd 17-19W-27ddc 17-20W-29cdd 36abb 18-16W-16bbb	19bba 23aaa 23add 28cdc 18-17W-11cda	14cdc 18ecc 27aba 28bcc 34bca	18-18W- 9aab 25bcb 26dda 28aac 30ddc	31dda 34bbb 36daa 18-19W-21cdb 27aaa

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7.5		
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340 340 340 270 420	52 480 160 350 100	84 46 48 48
15 14 5.8 5.8 7.1	1.0 88 1.9 9.7 1.7	2.7 8.8 4.4 3.6
ಬೆ ಬೆ ಬೆ ಬೆ ಬೆ	70 6.1. 6. 6. 6. 6.	0.5.7.4 0.4.2.6.
22 142 142 143 143	340 100 20 90 90	93 160 250 120
63 79 89 89 89	220 74 19 77 28	120 200 200 170
322 310 356 277 424	361 316 292 312 299	278 327 442 317
19 28 36 20	$\frac{450}{32}$ $\frac{32}{59}$ $\frac{21}{140}$	190 310 420 260
112 122 10	4.8 16 17 11 13	8.9 5.1 7.7
120 120 110 89 150	13 170 37 120 20	19 10 6.6 6.6
0.00.00.00.00	88888	6.666
.20 .20 .20 .10	.55 .53 .06 .18	.71 .15 .22 3.0
33300 33300 33300	$\begin{array}{c} 8.0 \\ 20 \\ 11 \\ 33 \\ 8.0 \end{array}$	11 7.0 14 9.5
463 465 479 418 535	1,220 657 309 469 448	591 861 1,130 745
12.0 13.0 12.0 21.0	14.0 12.0 15.5 12.0 16.0	16.0 13.5 19.0
8-11-60 8-11-60 8-11-60 8-22-60 2- 6-61	10-11-60 10- 7-60 5-17-60 8-15-60 5-16-60	$\begin{array}{c} 5.16-60 \\ 10-11-60 \\ 5-13-60 \\ 5-13-60 \end{array}$
	Kd Qv,Kd Kd Qt,Kd Kd	Kd Kd Kd
44 63 54 stream 47	214 80 185 140 390	290 284 360 410
144 44 45 45 45	46 48 49 50	223 24 24 27
18-20W-14ccb 15ccb 19acc 19ccb 20dca	36abb 19-16W-23ddb 19-17W-27add 19-18W- 3aca 27ccb	19-19W-24ccb 19-20W- 4bbb 17ddc 26baa

analyzed by Kansas State Department of Health

One milligram per liter is equivalent to 8.33 pounds of substance per million gallons of water.
 Kd, Dakota Formation; Qt, terrace deposits; Qu, undifferentiated Pleistocene deposits; Qv, valley-fill deposits.
 Sodium and potassium expressed as sodium.

Thus, a storage coefficient of 0.15 was assigned to the valley-fill deposits.

CHEMICAL CHARACTER OF GROUND WATER

Chemical character of ground water in Rush County is indicated by analyses of samples of water collected from 51 wells and a seep. The analyses of two water samples from Walnut Creek also are included. Results of the chemical analyses are given in milligrams per liter in table 10, but many of the data were examined using units of milliequivalents per liter. Milligrams per liter can be converted to milliequivalents per liter (meq/l) by multiplying milligrams per liter by the factors given in table 11.

TABLE 11.—Factors for converting milligrams per liter to milliequivalents per liter.

Mineral constituents	Chemical symbol	Multiply by	
Cations			
Calcium	Ca++	0.04990	
Magnesium		.08226	
Sodium	Na+	.04350	
Potassium	K+	.02558	
Anions			
Carbonate	CO ₃	.03333	
Bicarbonate	HCO ₃ -	.01639	
Sulfate	SO ₄	.02082	
Chloride	Cl	.02821	
Fluoride	F-	.05264	
Nitrate	NO ₃ -	.01613	

Water from the seep, which is near an abandoned oil well, had unusually large concentrations of dissolved constituents. The sample is not considered to be representative of water from aquifers of Cretaceous or younger age: therefore, the analysis is shown only in table 10 and on figures 11-13, and is not included in the discussion of ranges in concentration of chemical constituents in ground water.

Chemical Constituents in Relation to Use

The following discussion of the chemical constituents of ground water in relation to use has been adapted in part from publications of the U.S. Geological Survey and the State Geological Survey of Kansas. The recommended maximum concentration of chemical constituents in water used for domestic purposes (U.S. Public Health Service, 1962) are summarized in table 12.

Dissolved solids.—When water evaporates, the residue consists of the mineral constituents listed in table 10 and may contain some organic matter and water of crystallization. Water containing more than 1,000 mg/l of dissolved solids is generally objection-

Table 12.—Recommended maximum concentrations of chemical constituents in water used for domestic purposes (adapted from U.S. Public Health Service, 1962).

Constituent	Concentration (milligrams per liter
Chloride (Cl)	250
Fluoride (F)	
Iron (Fe)	3
Manganese (Mn)	
Nitrate (NO ₃)	45
Sulfate (SO ₄)	250
Dissolved solids	

able for domestic and municipal use; however, in various sections of the country, and in Rush County, water that contains more than 1,000 mg/l dissolved solids is used with no apparent adverse effect. More than 60 percent of the water samples from the Dakota Formation in Rush County contained more than 1,000 mg/l dissolved solids. The dissolved-solids concentration of water samples ranged from 309 to 3,660 mg/l. Of 53 samples, 19 contained less than 500 mg/l, 14 contained 500 to 1,000 mg/l, and 20 contained 1,000 or more mg/l dissolved solids.

Hardness.—Hardness in water may be recognized by the increased quantity of soap required to produce lather; it is a major contributor to scale formation in boilers, radiators, water heaters, and pipes. Akaline earths, mainly calcium and magnesium, commonly cause hardness in water, but iron and manganese may also contribute when they are present in high concentrations.

Carbonate hardness, or temporary hardness, is that part of the hardness that is equivalent to the carbonate and bicarbonate present in water; it can be removed for household use almost completely by boiling the water. Noncarbonate hardness, or permanent hardness, is the remainder of the hardness; it cannot be removed by boiling.

Water that has 60 mg/l or less of hardness is usually considered to be soft and suitable for most purposes without further softening. Moderate hardness, which ranges from 61 to 120 mg/l, does not seriously interfere with the use of water for most purposes, but it does increase the amount of soap required. Water with hardness ranging from 121 to 180 mg/l is considered to be hard, and laundries and industries may profitably soften such supplies. Water with hardness of more than 180 mg/l is very hard and generally requires softening before use. The preceding ranges in hardness are from Durfor and Becker (1964).

The calcium and magnesium hardness of water samples collected in Rush County ranged from 24 to 490 mg/l. Of 53 samples, 26 were very hard, 2 were hard, 14 were moderately hard, and 11 were soft.

Iron.—Some iron is present in most ground water. The quantity present may vary significantly from place to place, even within the same aquifer. When water contains 0.1 to 0.3 mg/l or more of iron in solution, the iron upon oxidation may precipitate as a red sediment. Iron may give a disagreeable taste to water, it may stain clothing, cooking utensils, and plumbing fixtures, and it may be objectionable in the preparation of foods and beverages. Iron can be removed from most water by aeration and filtration, but chemical treatment also may be required.

The concentration of total iron in water samples collected from wells in Rush County ranged from 0.01 to 3.0 mg/l. The sample from Walnut Creek near the east side of the county had a concentration of 3.2 mg/l. Of 53 samples, 28 contained less than 0.30 mg/l, and 25 contained 0.30 mg/l or more of iron.

Manganese.—Manganese has properties similar to iron and is considered with iron when evaluating the usefulness of water. Only 10 samples of water collected in Rush County contained measurable amounts of manganese and the concentration present ranged from 0.05 to 0.70 mg/l.

Chloride.—Chloride is widely distributed in nature; it is an abundant constituent of sea water, oil-field brines, and streams and lakes in arid or semiarid regions. In ground water the concentration of chloride may range from a few to more than 100,000 mg/l. Small quantities of chloride may be dissolved from most rock materials. Chloride has little effect on the suitability of water for ordinary use unless concentrations are high enough to make the water impotable or corrosive. The removal of chloride from water is costly and uneconomic for most present-day uses. Water containing 200 to 300 mg/l of chloride, and an equivalent amount of sodium, may have a salty taste.

The chloride content of water samples collected in Rush County ranged from 14 to 1,800 mg/l. Of 53 samples, 35 contained 250 mg/l or less, and 18 contained more than 250 mg/l of chloride.

Fluoride.—Fluoride generally is present in ground water in small quantities. Knowledge of the concentration of fluoride in water used by children is quite important, as concentrations of about 1 mg/l in drinking water lessens the incidence of tooth decay (Dean and others, 1941). The use of water containing more than 1.5 mg/l fluoride during the time of formation of permanent teeth may cause mottled tooth enamel (Dean, 1936).

The fluoride concentration of samples of water from Rush County ranged from 0.1 to 5.6 mg/l. Of 53 samples, 25 contained 1.2 mg/l or less and 28 contained more than 1.2 mg/l of fluoride.

Sulfate.—Sulfate in ground water is derived chiefly

from solution of gypsum and anhydrite, or by oxidation of sulfides of iron. Water containing more than 250 mg/l sulfate will have a laxative effect on some of those who drink it.

The sulfate content of water samples in Rush County ranged from 17 to 450 mg/l. Of 53 samples, 48 contained less than 250 mg/l and 5 contained more than 250 mg/l of sulfate.

Sodium.—Sodium commonly is present in ground water and its concentration may range from a few milligrams per liter in water in some aquifers to more

than 100,000 mg/l in brines. Water containing 130 to 200 mg/l of sodium, and an equivalent amount of chloride, may have a salty taste.

A large percentage concentration of sodium in water to be used for irrigation is undesirable because of its effect on soils. The concentration of sodium in water in relation to its use for irrigation is discussed in a later section.

The concentration of sodium in water samples in Rush County ranged from 13 to 1,300 mg/l. Of 53 samples, 29 contained less than 250 mg/l and 24 con-

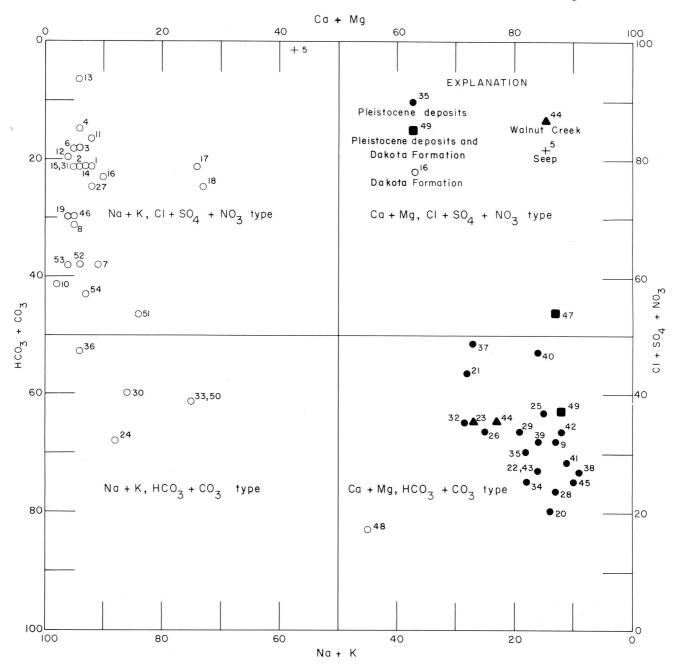


Figure 11.—Modified Piper diagram of percent milliequivalents per liter of cations and anions for water samples. Numbers by symbols are sample-identification numbers from table 10.

tained more than 250 mg/l of sodium.

Nitrate.—In 1954, a physician in Iowa reported cyanosis, or methemeglobinemia, in infants whose feeding formulas were mixed with water containing high concentrations of nitrate (Comly, 1945). This so-called blue-baby disease appears to be a hazard when waters containing more than 45 mg/l of nitrate are used for infant feeding (Hem, 1970). Older children and adults seemingly are not harmed by moderate concentrations of nitrate in water. Boiling or softening of water for household use will not remove or decrease the nitrate content.

The concentration of nitrate in water samples collected from Rush County ranged from 0.4 to 88 mg/l. Only one sample exceeded 45 mg/l nitrate; all other

samples contained less than 35 mg/l. In a recent study by the county agent, 748 wells were sampled and 201 were found to contain water with 45 mg/l or more nitrate (E. L. VanMeter, written commun., 1971). The high concentrations of nitrate were commonly found to result from surface contamination due to improper location and sealing of the well.

Silica.—Silica is present in most water and may be deposited with other scale-forming minerals in steam boilers, but it has little effect on the use of water for other purposes.

The concentration of silica in water samples in Rush County ranged from 5.0 to 52 mg/l. Of 53 samples, 31 contained 20 mg/l or less and 22 contained more than 20 mg/l of silica.

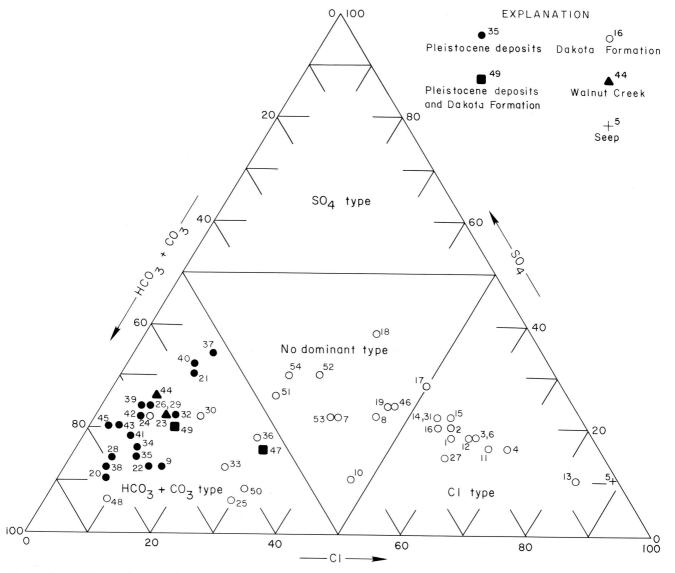


Figure 12.—Trilinear diagram of percent milliequivalents per liter of anions for water samples. Numbers by symbols are sample-identification numbers from table 10.

Chemical Constituents in Relation to Location and Aquifer

Samples of ground water were collected from 51 wells in Rush County; 30 samples were from the Dakota Formation of Cretaceous age, 19 samples were from aquifers of Pleistocene age, and two samples were from Pleistocene and Dakota deposits. Surfacewater samples were collected from Walnut Creek near the east and west county lines.

AQUIFER DELINEATION AND WATER TYPE

Figures 11, 12, and 13 are diagrams modified from those proposed by Piper (1944), in which milliequiv-

alents per liter of cations and (or) anions are plotted as percentages of total milliequivalents per liter of cations or anions. The diagrams are useful for comparing the chemical relations of water from different sources and for classifying the water by chemical type. For example, by examination of the values plotted on the diagrams, water from the Dakota Formation can be distinguished readily from water from the Pleistocene deposits. Most of the analyses for water samples from the Dakota indicate water of the sodium chloride type and most of those for samples from the Pleistocene deposits indicate water of the calcium bicarbonate type.

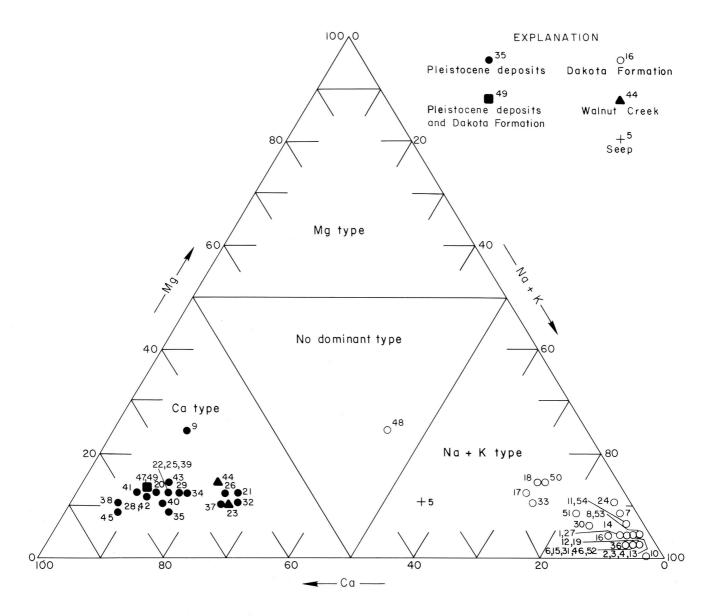


FIGURE 13.—Trilinear diagram of percent milliequivalents per liter of cations for water samples. Numbers by symbols are sample-identification numbers from table 10.

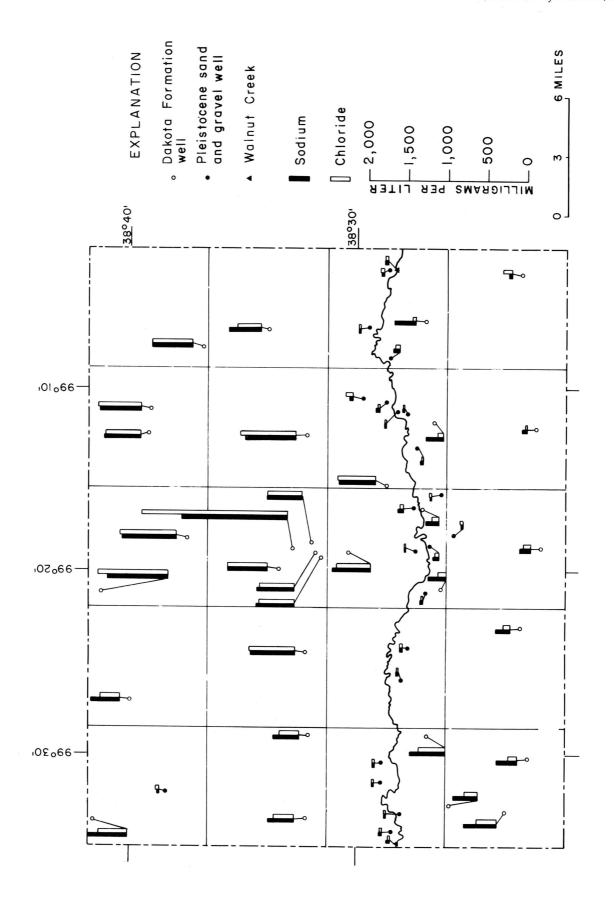


FIGURE 14.—Concentrations of sodium and chloride in water samples.

SODIUM AND CHLORIDE

The large range in concentration of sodium and of chloride by location and aquifer is illustrated on figure 14. A significant decrease in chloride concentration relative to sodium concentration, as well as a noticeable decrease in total sodium and chloride, in water from the Dakota is apparent on the south side of Walnut Creek in Rs.16-19 W. The reduction in sodium and chloride concentration may have resulted from recharge and dilution of water in the Dakota by water from the sand and gravel of Pleistocene age where the two aquifers are in contact in the valley of Walnut Creek. The area of contact begins somewhere in R.19 W. and continues eastward through the county.

HARDNESS AS CACO3 AND SULFATE

The locations where water samples were collected and the range in concentration of hardness and of sulfate in the water are shown on figure 15. In general, Pleistocene deposits contain water with higher concentrations of hardness and lower concentrations of sulfate than the Dakota. Water from the Dakota has a wide range in concentration of hardness and of sulfate. Figure 15 also indicates a possibility of some recharge and dilution of water in the Dakota by water from Pleistocene deposits, as there is some decrease in sulfate concentration in water from the Dakota on the south side of Walnut Creek.

FLUORIDE, IRON, AND MANGANESE

The locations where water samples were collected and the fluoride, iron, and manganese concentrations in the water are shown on figure 16. The large fluoride concentrations in all but three of the Dakota water samples and the smaller concentrations of fluoride in the water from the Pleistocene aguifers are readily apparent. Concentrations of iron differ greatly in water samples from both aquifers, and no pattern for amount of iron or location of sample is apparent. Manganese was found in 10 of the samples analyzed. Two samples of Dakota water contained small concentrations of manganese, and the sample from Walnut Creek on the east side of the county contained a relatively large concentration. Manganese was contained in seven other samples from the Pleistocene aquifer, but no pattern of concentration or location of samples is discernable.

SILICA

Inspection of the chemical analyses of water samples from Rush County reveals large differences in concentration of silica in water from the Dakota and in water from the Pleistocene aquifers. None of the 30 samples of water from the Dakota had concentrations that exceeded 14 mg/l of silica, and no samples of water from the Pleistocene deposits had concentrations less than 26 mg/l of silica. Thus, a concentration of 14 mg/l or less of silica for a water sample from an unknown aquifer would indicate a probable source of Dakota Formation for that sample. Figure 17 shows the locations and concentrations of silica for the water samples.

Chemical Constituents in Relation to Use of Water for Irrigation

The following discussion of the suitability of water for irrigation is adapted from Agricultural Handbook 60 of the U.S. Department of Agriculture (U.S. Salinity Laboratory Staff, 1954).

The development and maintenance of successful irrigation projects involves controlling the salinity and alkalinity of soils as well as supplying irrigation water to the land. Irrigation water quality, irrigation practices, and soil-drainage conditions are involved in saline and alkali soil control. Inadequate soil drainage and soil management or improper irrigation practices may cause soil to become unproductive through the accumulation of excess soluble salts or exchangeable sodium.

In areas where rainfall is sufficient and soil conditions are ideal, the soluble salts present in the soil are carried downward by percolating water and ultimately reach the water table. The process of dissolving and transporting soluble salts by the downward movement of water through the soil is termed leaching. The amount of water applied to the soil must be in excess of the amount needed by plants, or water will not percolate below the root zone, and mineral matter will accumulate at that point. Zones of impermeable soil near the surface can retard the downward movement of water, the soil may become waterlogged, and salts may be deposited. Unless soil drainage is adequate, leaching operations may not be successful, because leaching requires the passage of water through and away from the root zone.

The chemical characteristics of water that are the most important in determining its quality for irrigation are: (1) total concentration of soluble salts, (2) relative proportion of sodium to other principal cations (calcium, magnesium, and potassium), (3) concentration of boron or other toxic elements, and (4) under some conditions, the concentration of bicarbonate as related to the concentration of calcium plus magnesium.

Diagnosis and classification of the total concentra-

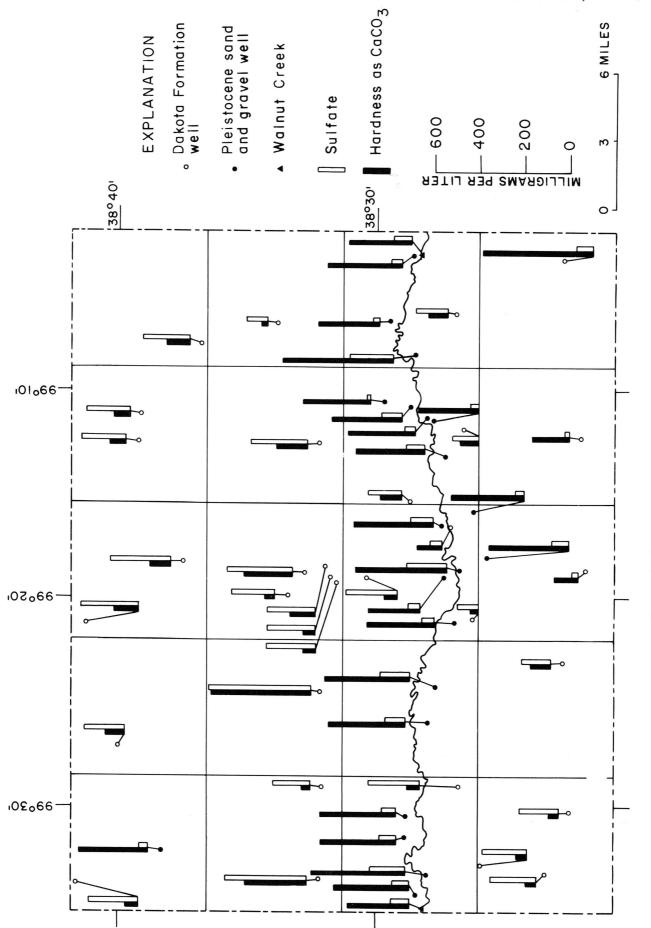
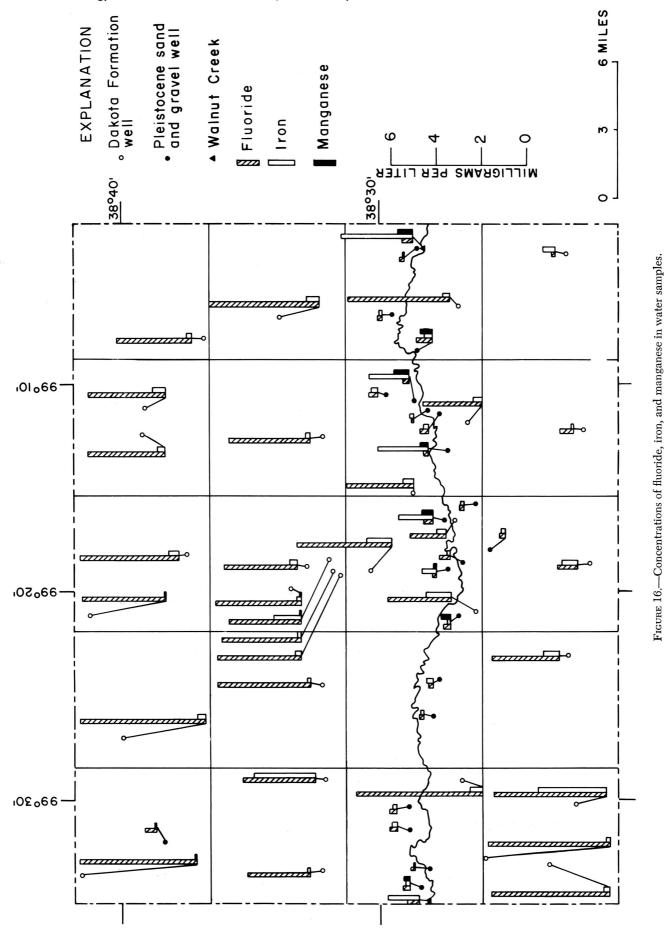
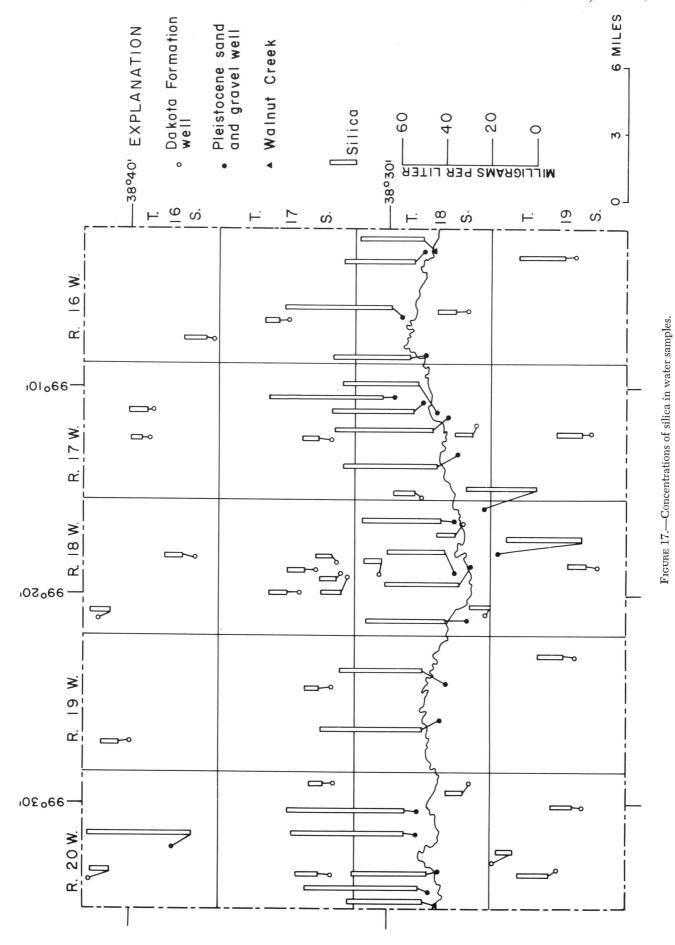


FIGURE 15.—Concentrations of hardness and sulfate in water samples.





tion of soluble salts in irrigation water can be expressed in terms of electrical conductivity (specific conductance). Electrical conductivity is a measure of the capacity of ionized inorganic salts in solution to conduct an electrical current; it is generally expressed in micromhos per centimeter at 25°C. The electrical conductivity can be determined accurately in the laboratory, or an approximation of the electrical conductivity can be obtained by multiplying the total milliequivalents per liter of calcium, magnesium, sodium, and potassium by 100, or by dividing the dissolved solids in milligrams per liter by a factor that averages 0.64. However, in Rush County, only eight of the 53 analyses would be close to 0.64. The factor for most of the analyses in Rush County is about 0.60. Water with conductivity values below 750 micromhos per centimeter is satisfactory for irrigation insofar as salt content is concerned, although crops sensitive to salt may be adversely affected by irrigation water with conductivity values in the range 250 to 750 micromhos per centimeter. Water in the range 750 to 2,250 micromhos per centimeter is widely used, and satisfactory crop growth is obtained under favorable soildrainage conditions and good management, but saline soil conditions will develop when leaching is incomplete. Very few instances are known where water with conductivity values greater than 2,250 micromhos per centimeter has been used successfully. The use of copious quantities of such water on soils and subsoils with excellent drainage may allow the more salttolerant crops to be grown.

The alkali hazard of an irrigation water is determined by the absolute and relative concentrations of sodium, potassium, calcium, and magnesium. In the past the relative proportion of sodium to the other principal cations in irrigation water generally was expressed as the percentage of sodium, expressed as milliequivalents, among the principal cations (referred to as the percent sodium). The sodium-adsorption ratio (SAR), used to express the relative activity of sodium ions in exchange reactions with soil, is a better measure of suitability of water for irrigation with respect to the sodium (alkali) hazard. The sodium-adsorption ratio may be determined by the formula

$$SAR = \frac{Na^{+1}}{\sqrt{\frac{Ca^{+2} + Mg^{+2}}{2}}}$$

with ionic concentrations expressed in milliequivalents per liter. If the sodium-adsorption ratio and the electrical conductivity of a water are known, the suitability of the water for irrigation can be determined by plotting these values on the diagram shown on figure 18.

The significance and interpretation of the diagram

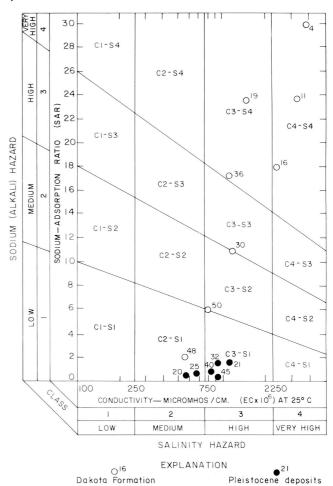


Figure 18.—Diagram showing suitability of water for irrigation. Numbers by symbols are sample-identification numbers from table 10.

for the classification of irrigation waters is as follows:

Low-salinity water (C1) can be used for irrigation with most crops on most soils with small probability that soil salinity will develop. A minimum of leaching is required; minimum leaching generally occurs under normal irrigation practices, except in soils of very low permeability.

Medium-salinity water (C2) can be used if moderate leaching occurs. Moderately salt-tolerant plants generally can be grown without special salinity-control practices.

High-salinity water (C3) cannot be used on soils of restricted drainage. Special management for salinity control may be required and plants with good salt tolerance should be grown, even on soils with adequate drainage.

Very high salinity water (C4) may be used occasionally under special circumstances. The soil must be permeable and have excellent drainage, irrigation water must be applied in excess to provide maximum leaching, and extremely salt-tolerant crops must be selected.

Low-sodium water (S1) can be used on most soils with scant danger of the development of harmful levels of exchangeable sodium.

Medium-sodium water (S2) present an alkali hazard in fine-textured soils under low-leaching conditions, unless gypsum is present in the soil. This water is usuable on coarse-textured or organic soils of high permeability.

High-sodium water (S3) produces harmful levels of exchangeable sodium in most soils and requires special soil management for successful use. The soil must have good drainage, a high leaching rate, and organic matter additions. Chemical amendments such as calcium chloride, gypsum, sulfur, sulfuric acid, crushed limestone, and others may be required for replacement of exchangeable sodium. Harmful levels of exchangeable sodium may not develop in gypsiferous soils.

Very high sodium water (S4) generally is unsatisfactory for irrigation except at low to medium salinity, where solution of calcium from the soil or the use of a chemical amendment may make these waters usable.

All water samples from the Pleistocene aquifers in Rush County have a low alkali hazard and medium to high salinity hazard. Irrigation is successful in Rush County with water of this quality, which implies acceptable soil drainage with adequate leaching and good soil-management practices. Water from the Dakota is much less acceptable for irrigation. Water from the Dakota cannot be used to irrigate lawns and domestic gardens satisfactorily in much of the county. If the reader will note the position of Dakota water samples on figure 18, he will have a clearer understanding of why water from most of the wells in the Dakota cannot be used for irrigation, even for a garden plot, without careful management.

An appraisal of the quality of irrigation water should be concerned first with the determination of the salinity and alkali hazards by reference to figure 18. Consideration then may be given to independent characteristics, such as boron, other toxic elements, and bicarbonate, any one of which may change the quality rating. Soil-drainage characteristics and management, as well as water quality, must be considered also before water-use recommendations are made. The interested reader is referred to Agricultural Handbook 60 (U.S. Salinity Laboratory Staff, 1954) for a more complete discussion of the suitability of water for irrigation.

Sanitary Quality

The analyses of water listed in table 10 show only the amounts of dissolved mineral matter in the water and do not indicate the sanitary quality of the water. However, high concentrations of certain minerals, such as nitrate or chloride, may indicate pollution of the water supply.

The municipal supplies in Rush County are obtained from properly constructed wells, which meet the requirements of and are periodically examined by the Division of Environmental Health of the Kansas State Department of Health. Many of the people of the county, however, have private water supplies, and precautions should be taken to protect these supplies from pollution. Possible sources of pollution, such as barnyards, privies, and cesspools, should be as far as possible from well locations, and well casings should be sealed to a level below the water table. In general, dug wells are more vulnerable to contamination from the surface than drilled and cased wells, because they often are not effectively cased or sealed at the surface.

UTILIZATION OF GROUND WATER

Ground water in Rush County is used for domestic, stock, municipal, industrial, and irrigation supplies. The largest use is for irrigation of about 10,000 acres in Walnut Creek valley. Table 13 contains information pertaining to 222 wells in Rush County. The number of wells given according to principal use is: 60 domestic and stock wells, 10 public supply wells, and 152 irrigation wells.

SUMMARY

Ground water in Rush County is obtained principally from deposits of Pleistocene age and from the Dakota Formation of Cretaceous age. Maximum yields from irrigation wells tapping Pleistocene deposits in Walnut Creek valley have been measured as 1,200 gpm and reported as 1,500 gpm. The maximum reported yield from the Dakota was 500 gpm from a municipal well at La Crosse.

Water in storage in the Pleistocene deposits in Walnut Creek valley provides water for intensive irrigation. Recharge to the Pleistocene aquifer in the valley occurs quickly and in large amounts during periods of high flow in Walnut Creek. About 50,000 acre-feet of water may have been recharged in September 1959 by the maximum flood of record on Walnut Creek.

Water from Pleistocene deposits is hard to very hard, but otherwise is suitable for most uses. Dissolved-solids concentrations of water samples from Pleistocene deposits ranged from 369 to 789 mg/l. Water from the Dakota Formation is variable in chemical quality and generally is considered to be marginal for domestic use and for irrigation. Dissolved-solids concentrations of water samples from the Dakota ranged from 309 to 3,660 mg/l.

Table 13.—Records of wells.

Altitude of land surface above mean	ure- sea level (gallons ment (feet) per minute)	0-60 1,911 0-60 1,877 0-60 1,947 2,005			3,000					
Depth Date below of land meas		S 8.7 10-60 S 3.8 10-60 S 12.3 10-60 D, S D, S D, S	7. 6. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.	27.5		27.5 7.0 10.1 16.3 22.6 8.6 7.0 7.4	6.72 6.70 6.70 6.88 6.88 7.70 6.70 7.70 7.70 7.70 7.70 7.70 7.70	6. 0. 1. 0. 1. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.	2.7. 10. 7. 10. 10. 10. 10. 10. 10. 10. 10. 10. 10	6. 7. 10. 7. 10. 10. 10. 10. 10. 10. 10. 10. 10. 10
unit Method of lift,7	type or power ⁸	SH Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z Z	T, E Cy, W Cy, E, W Cy, W Cy, W		Cy, H, W Cy, E, Cy, E Cy, W Cy, W	ÇÇÇ, W ÇÇÇÇ, W ÇÇÇÇ, W Ç, E, W	MANA HANA HANA SÖĞĞĞ ÜĞĞĞĞĞ ÖĞĞĞĞĞ SÜĞĞĞĞĞ ÜĞĞĞĞĞĞĞ			
ter-b	materials source	Sd, St Sd, St Sh, St Ss, Kd Kd Ss Kd	Ss Kd Sd, St Qt Ss Kd Sd, St Qu, F Sd, Gr, St Qt	St	Ss Kd Sd, St Qv Sd, St Qv	St St St St St		**************************************	% % %% % % % % % % % % % % % % % % % %	K Č K K KKČČČ ČKKK K KKČČ KK
1	ing4	ъ 6161 Ощ ОО	36 2 36 36 8 8 8	48 R 5 G	72 R	36 38 38 38 38 38 38 38 38 38 38 38 38 38				
Depth of well3	(feet)	10 9 21 266 R 370 R	320 R 10 230 R 18 33	22 16 209 R	15 23	15 23 28 12 12 27 30 30	### ### ### ### ### ### #### #### ######	### ### ### ### ######################	7.12 821704 821704 821704 8	21 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2
Owner	or user	J. Kisner E. Hopkins Jacob Wegele Jeff Allen Mary Basgall	St. Mary Church J. Werth C. N. Bieker W. Kreutzer M. Bahr	John P. Enslinger Schuckman Noel Mellick	B. Legleiter W. R. Jefferies	B. Legleiter W. R. Jefferies W. R. Jefferies Leonard Kreutzer G. E. Schutte Albert Zimmerman Mrs. M. B. Close John Oelkers	B. Legeiter W. R. Jefferies W. R. Jefferies Leonard Kreutzer G. E. Schutte Albert Zimmerman Mrs. M. B. Close John Oelkers Rudolph Schoendaller J. H. Kreutzer L. Mosher Harry Saunders Estate M. F. Littler	B. Legleiter W. R. Jefferies W. R. Jefferies Leonard Kreutzer G. E. Schutte Albert Zimmerman Mrs. M. B. Close John Oelkers Rudolph Schoendalle J. H. Kreutzer L. Mosher Harry Saunders Estat M. F. Littler Cemetery F. M. Hardwick G. J. Huber F. Komerak A. H. Foster	B. Legleiter W. R. Jefferies W. R. Jefferies Leonard Kreutzer G. E. Schutte Albert Zimmerman Mrs. M. B. Glose John Oelkers Rudolph Schoendalle J. H. Kreutzer L. Mosher Harry Saunders Estat M. F. Littler Cemetery F. M. Hardwick G. J. Huber F. Komerak A. H. Foster Fred Black V. Jarvis Alvin Janke A. Ochs Orland Scheuerman	B. Legleiter W. R., Jefferies W. R., Jefferies Leonard Kreutzer G. E. Schutte Albert Zimmerman Mrs. M. B. Close John Oelkers G. H. Kreutzer L. Mosher Harry Saunders Estat M. F. Littler Cemetery F. M. Hardwick G. J. Huber F. Komerak A. H. Foster Fred Black V. Jarvis A. H. Foster Fred Black V. Jarvis A. H. Foster Fred Black V. Jarvis Orland Scheuerman Emma Schwartzkopf School District W. Weber H. J. Kleihege L. V. Mellick
Well	number ^{1,2}	16-16W- 8bcc 10ddd 28aba 32ccb* 16-17W-16dcd*	22aaa* 16-18W- 3cdc 6dac* 7bcc 9bad	18ccd 20bbb 97,41*	27cdd* 28aba 29add	2/cdd* 28aba 29add 30ccd 16-19W- 6daa 17bab* 19ddd	27cdd* 28aba 29add 30cd 16-19W- 6daa 11bba 17bab* 19ddd 25abb 25abb 27ada 27bcb	2/cdd*28aba 28aba 30ccd 30ccd 16-19W- 6daa 17bab*19ddd 25abb 25dad 27ada 27bcb 29aaa 30bbb 324bcb 34bcb 35aba 16-20W- 2dac	27.cdd* 28aba 28aba 29add 30ccd 16-19W- 6daa 117bab* 19ddd 27ada 27ada 27ada 27ada 27ada 27abb 28aaa 30bbb 32dbd 34bcb 334bcb 35aba 16-20W- 2dac 5bd* 6dcc 5bd*	

Table 13.—Records of wells—continued.

	e of face face Yield³) bei	©#100	9 4 660	2 9 1,080 8 650 B		5 1,500 R 7 700			1,000 R 800 R 550 R 1,250 R 670	2 720 2 100 R	9 8 900 E		800	2 800 E 2 1,350 R	1,100 R 5 650 R		5 1,400 R 4 3	98 890	002
	Altitude of land surface above mean	sea level (feet)	2,134 2,175 2,175	2,13 1,94		1,935 1,935 1,933				1,947 1,947 1,947 1,940 1,950	1,932 1,942 1,926	1,928 1,928		2,043 2,005 1,958	1,952	1,953	1,949	1,955 1,954 1,958	1,956	1,961
Water level	Date of meas-		09-6 09-6	2-60	5-60		5-60 5-60	и 1900 1900 1900 1900 1900 1900 1900 190	5-60	900000 9000000000000000000000000000000	7 2 60 7 60 7 60	2-60	5-60 5-60	9-60 5-60	ж 75 60 60 60 60	5-60 5-60	2-60	7,4,7 9,60 9,60	24.50 90.50	100 100 100 100 100 100 100 100 100 100
	Depth below	surface ³ (feet)	16.3 9.5	16.9	16.9 5.9	10.9 9.1	9.9	16.3 12.6	12.3	16.5 15.2 13.7 11.6	7.6 18.3 7.6	6.4	9.9 15.4	$\frac{23.7}{19.6}$	15.6 14.6	15.1	16.8	17.0 15.6 18.3	16.2	19.5
		Use	S PS D, S	D, S			II	H H 1	I	нини	I I PS	нн	I I C	L'S.I	ннь		нн			, ,,
		type of powers	Cy, H	Çỳ, ⊔	HHE	T,T	$_{ m T,T}$	⊢⊢	∐	, , , , , , , , , , , , , , , , , , ,	HHH	$_{ m T,LPG}^{ m T}$	H,T,	Çý, Cy, W≅	$_{\mathrm{T}}^{\mathrm{T}}$	$^{1,1}_{\mathrm{T}}$	T, T	$_{\mathrm{T,T}}^{\mathrm{T,T}}$	$_{T}^{\mathrm{L}}$	T,E
	Principal water-bearing unit	Geologic source ⁶		Кd Qv	òòò	\$\$\$	ÕÕ	ÒŎ	Ò	<u> </u>	òòò	ÒÒ			8	\$\$\$	òò	<u> </u>	òč	ŠÕ
	Principal wa	Character of material ⁵	Sd, Gr, St Sd, St Ss	Ss Sd, Gr	Sd, Gr			Sd, Gr		Sd, Gr Sd, Gr Sd, Gr Sd, Gr	Sd, Gr Sd, Gr Sd, Gr		Sd, Gr Sd, Gr	Sd, Gr, St Sd, Gr	Sd, Gr			Sd, Gr Sd, Gr Sd, Gr	Sd, Gr	
	type	cas- ing4	ßщ	I	Ŋ		1			HH	S					ל	П	S		Ι
	Diam-	well (inches)	36	16	16	22	24	19		19 18 16 16	42	16 19	15	36 19	16 16	16 16 16	16 16	19 16	16	19
	Depth	well ³ (feet)	34 11 285 R		79 73	77 R 81 R	84 R 88 R	86 78 R	73	85 71 78 78 78 78	79 62 75	87 R 78 R	48 33 168 B	33 33 86 R	79 70 78	712	51	89 79 R 71	70 85 B	76
		Owner or user	V. Huxol School District O. S. Bellport	H. L. Baker John Luft	I. B. Pearson Willis C. Mohr	Harry Herbel Ed Mohr	Morris Brack do	H. E. Glantz Adolph Schroat	op	Henry J. Pechanec do do Adam Huenergardt Lloyd Rodie	H. E. Glantz Francis Schneider City of Otis	Joe Koochel do	Elmer Brack Glenn Moore	Mrs. Mary Ficken C. H. Hardy	S. A. Crotinger Andy Schuetz	D. r. rechance do W. R. Pechanec	Henry J. Pechanec B. F. Pechanec	Robert Kraisinger Lester Dirks do	Jerry Oborny Louis F. Oborny	Jean Oborny
		Well number ^{1,2}	1ddc 27add 29cdd*	36abb* 18-16W- 8ddc	9ccd 14bcc	15cbc 15cbc 15daa	16adc 16bac	16bbb 16dbd 1,	17abb	18acb 18cbd 18cdc 18dab 19bba◆	21aab 22cad 23aaa*	23bbb 23bca	24ccd 25bbc 98odo	18-17W- 6dda 11cda*	12cdc 13bbc	13ccb 13ccc	13dab 13dec	14abd 14adb 14bcc	18-17W-14cdc* 15adc	15dca

350 R 700 E 1,000 R	600 R	1,200 R		800 R	1,200 R 1,000 R	650 R	1,200 R 580	550 R	740			, ,	1,100 K 800 R	1,100 R 1,000 975	1,200 R 600 R 545	700 R	1,200 R 1,000 R	800 R 300 R	1,000 R	1,100 R 1,000 R
1,978 1,983 1,980	1,968 $1,972$	1,971	1,963 1,965 1,968	1,965	1,967 1,967 1,965	1,964	1,956 1,956 1,972	1,965 1,964 1,968	1,968 1,969	1,972 1,978 1,979	2,005 2,011	2,092 1,995	1,991 1,987	1,986 1,982 1,983 1,986 1,984	1,990 1,984 1,991	2,005 1,995	1,996 2,000	1,996 1,996 1,996	2,003 1,999 2,002	2,005 2,005 1,009
ကုက်က ဝဓိဓိဓိ	2 2 2 2 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3 3	5-60	77 77 79 99 99 99 99	77 77 77 90 90 90 90	77 77 900 900 900	ич 60 60	м 9 9 9 9 9 9 9 9	7 7 9 9 9 9 9	7 9 9 9	х 5-60 5-60 60	09-6	8-60 9-60 9-60	4-60 60 60	44444 999999	자 자 소 아 60 아 60	2-60	77.7 60 60 60	2-90 2-90 8-90 8-90	න්නන් මෙලිලි මෙලිලි	2-60
24.8 23.0 21.3	18.7 19.2	20.2	20.2 21.2 20.8	20.4 19.9	21.7 20.0	21.0	16.4 18.3 23.4	20.2 16.1 19.9	17.1	21.2 22.0 22.2	11.0	9.8 20.4	24:2 24:8	22.5 21.9 21.9 21.8 21.1	23.7 19.2 22.8	22.6	21.6	21.0 19.6 20.1	22.3 18.1	22.7
ннн		н-		ннг		н		ннн	HH	ннн	D, S	D, S I	HH	нннн	ннн	J I	ннь	I PS		ч — —
HHH,	T, T	$_{\mathrm{T},\mathrm{T}}^{\mathrm{T}}$	LHH,	T,T T,T	- [- [-	T,T	T, T	T, T,	.Н.Н .Н.Н	$_{ m T}^{ m L,NG}$	Çỳ, Cỳ,≪	$C_{\rm y,W}^{\rm Cy,W}$	$_{ m T,T}^{ m T}$	ттт, татт т	T,T T	Д, ЯЕ	$_{\mathrm{T}}^{\mathrm{T,T}}$	$^{ m T}_{ m T,E}$	T,T T,T	JEH.
<u> </u>	ර්ර්	ŠŠ	, 2000	8	355	òċ	ර්ර්ර්ර්	ර්ර්ර්	; & &	<u> </u>	Kd Qv	P & &	8	దేదేదేదేదే •	8	Qv Qv	8	\$&&	òòò	\$\$\$
Sch Gr Sch Gr Sch Gr	Sd, Gr Gr Gr	Sd, Gr	Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg Sg S	Sd, Gr	Sch Sch Sch Sch Sch Sch Sch Sch Sch Sch	Sd, Gr	Sd. Gr. Sd. Gr. Gr. Sd. Gr. Gr.	Seg. Seg. Seg. Seg. Seg. Seg. Seg. Seg.	Sd, Gr Sd, Gr	Sd, Gr Sd, Gr Sd, Gr	Ss Sd, Gr, St	Ss Sd, Gr, St Sd, Gr	Sd, Gr Sd, Gr	Sd, Gr. Sd, Gr. Sd, Gr. Sd, Gr.	Sd, Gr Sd, Gr Sd, Gr	Ss Sd, Gr	Sd, Gr	Sd, Gr Sd, Gr	Sd, Gr Sd, Gr	
						-	. н		Ι	H	ပ	Ö	Ι		(S	н	-
19 19	91 13	16	19 19 14	19	16	17	16	19	16 19	16	9	9	16 19	19 19 16	19 16 14	16	19	10 10	16	18
74 70 66	74 R 65 R	88	72 70 R 64	70 R 69	01 h 72	70	57 71 59	52 54 54 54 54	99 63	63 75 R 69	175 R 32	265 R 26 66	73 73	71 81 R 67 72 R	71 R 54 56	187 K 72 R	69 R 67	83 R	65 50 50	68 R 80 R
	de Ralph Pivonka ed R. E. Gisiek		E. J. Pechanec do do do do		oc James Jecha bo do J. A. Vondracek	·	ob V. Kraisinger bd S. Vecesky ba* Ed Richter	ba Ernest Bucl bc Al Holopirek sa Ed Iuno						bc Joe Haberman cc Rudolph Oborny cb* Frank Oborny a C. O. Dighton lb Dr. J. H. Baker		da* Ted Barron ıb Frank Jira		ac J. Flamik dc B. L. Moore cc* City of Rush Center	d Ernest Hallett	
19add 19cbc 19ccc	20ade 20bed	20dcc 20ddb	21acb 21acb 21adc 21bbb	21bd 21cb	21cdc 21dbc 21ddc	22cdb	23abb 23bbd 27aba	27bba 28abc 28aca	28bbc 28bcc*	28cba 30abb 30bab	34bca* 18-18W- 2ada	9aab* 19ada 22ccd	23ccd 24bcc	24cbc 24dcc 25bcb * 26aba 26adb	26bab 26dbb 26dcb	26dda 27aab	27bac 27ccb	27 dac 27 ddc 28aac*	28ccd 28dbb	29acc 29acc 29bdb

Table 13.—Records of wells—concluded.

19 School Colored Colored			Depth	Diam-	Type	Principal water-bearing unit	bearing unit	Method		Wate Depth below	11 1	Altitude of land surface	
D. E. Balloy 69 19 Sci. Gr QV T. LPG W. L. Balloy 72 19 Sci. Gr QV T. LPG N. L. Balloy 72 19 Sci. Gr QV T. LPG N. C. Milling 150 R 16 Sci. Gr QV T. T. T N. C. Milling 550 R 16 Sci. Gr QV T. T I. V. P. March 55 16 Sci. Gr QV T. T I. V. Diverses 54 16 Sci. Gr QV T. T J. S. Oliverus 55 16 Sci. Gr QV T. T H. W. Grass Estate 56 16 Sci. Gr QV T. T J. S. Oliverus 55 16 Sci. Gr QV T. T I H. W. Grass Estate 56 18 Sci. Gr QV T. T I Coroga Kershner 55 18 Sci. Gr QV T. T I C. William 54 18 <th>er^{1,2}</th> <th>Owner or user</th> <th>of well³ (feet)</th> <th>eter of well (inches)</th> <th>of cas- ing4</th> <th>Character of material⁵</th> <th>Geologic source⁶</th> <th>of lift,7 type of power⁸</th> <th>Use</th> <th>land surface³ (feet)</th> <th>ابد ي</th> <th>above mean sea level (feet)</th> <th>Yield³ (gallons per minute)</th>	er ^{1,2}	Owner or user	of well ³ (feet)	eter of well (inches)	of cas- ing4	Character of material ⁵	Geologic source ⁶	of lift,7 type of power ⁸	Use	land surface ³ (feet)	ابد ي	above mean sea level (feet)	Yield ³ (gallons per minute)
Roy Button 150 R 19 Sci Gr Qv T, E D S W. C. Mattin 57 150 R 16 85 Gr Qv, T, T 1 W. C. Mattin 57 16 85 Gr Qv, T, E 1 T. Y. House Estate 59 16 85 Gr Qv, T, E 1 A do do 54 16 85 Gr Qv T, E 1 I. S. Oliverus 59 18 85 Gr Qv T, E 1 I. S. Oliverus 56 18 85 Gr Qv T, E 1 N. Kirby Caeorge Kershner 55 25 1 85 Gr Qv T, T 1 N. Kirby C. Williams 55 25 1 85 Gr Qv T, T 1 Dan W. Pallips 61 R 16 85 Gr Qv T, T 1 Licov Semma 58 16 85 Gr Qv T, T 1 Licov Semma 58 16	30adc 30bbc 30cac 30cdb 30ddc*	D. E. Bailey B. and K. Schneidemind W. L. Bailey Ken Button Henry Howe		19 19 19			<u> </u>	$egin{array}{c} \mathrm{T} \\ \mathrm{T,LPG} \\ \mathrm{T,T} \\ \mathrm{T,E} \end{array}$	нннн	16.7 19.5 22.3 25.0 27.2	77776 77776 60 60 60	44444 00000 70000 701000 7014 710	1,000 R 900 R 1,000 R 1,000 R 890
F. Coliverus	31bda 31dda* 32aba 33abb 34acb	Roy Button Mrs. E. Jilg W. C. Martin L. V. Reynolds T. T. House Estate	63 R 150 R 57 56 105 R	19		Sd, Gr Sd, Gr Sd, Gr Ss, Gr	Qv Qv Qv, Kd	ÇÇ, H, H, H, H, H, H, H, H, H, H, H, H, H,	D, S I I	33.5 22.8 21.0	5-60 5-60 6-60 6-60	2,024 2,007 2,005 2,004 1,997	1,000 R
George Kershner 55 19 1 Sd, Gr Qv T, T I N. Kirby C. Williams 49 Sd, Gr Qv T, TG I Andy Felder 57 Sd, Gr Qv T, TG I Andy Felder 57 Sd, Gr Qv T, T I Dan W. Phillips 44 48 Sd, Gr Qv T, T I Levo Seaman 58 16 Sd, Gr Qv T, T I Levo Seaman 58 16 Sd, Gr Qv T, T I Levo Seaman 58 16 Sd, Gr Qv T, T I Levo Seaman 54 19 Sd, Gr Qv T, E I Levo Seaman 54 19 Sd, Gr Qv T, E I Loyd West 54 19 Sd, Gr Qv T, T I H. O, Jones 65 19 Sd, Gr Qv T, T I	34bbb 34bbc 36dca* V-15ccc 15ccd	do do J. S. Oliverus H. W. Grass Estate do	00000000000000000000000000000000000000	16 16 14 18		SS	<u> </u>	HH HHHHHH HHHHH	нннн	21.1 19.6 26.2 22.2 19.9	6-60 6-60 6-60	2,000 1,999 2,024 2,031 2,029	400 R 400
Dan W. Phillips 44 48 Sd, Gr Sd, Gr Sd, Gr Dav West Qv TT T T I I I I I I I I I I I I I I I I I I I	18ccc 19abc 20abc 20bbc 20bbc	George Kershner T. Baldwin N. Kirby C. Williams Andy Felder	55 55 57 57	19 25	H	Sd, Gr Sd, Gr Sd, Gr Sd, Gr	<u> </u>	$\overset{\text{T}}{\overset{\text{T}}{\text{Y}}}\overset{\text{T}}{\overset{\text{T}}{\text{Y}}}$	нннн	25.56 25.66 25.25 25.25 25.35	09-9 09-9 09-9	2,048 2,043 2,039 2,041 2,041	400 R 500 R 450 R 500 R
H. J. Ford bones	20dbc 21acc 21cca 21cdb* 21dca	Dan W. Phillips H. W. Grass Estate Leroy Seaman J. H. Webs Lloyd West	44 61 R 60 R 58 60 R	48 16 16 19			\$\$\$\$\$	1,1 1,1 1	нннн	23.1 19.1 26.4 25.7 24.9	6-60 6-60 6-60 7-60	2,038 2,031 2,038 2,036 2,034	600 R 450
L. J. Ford 70 19 Sd, Gr Ov T, E 17 1 Frank Kershner 78 R Sd, Gr Ov T, E 1 1 John Weitzel Vera Bitters 77 19 Sd, Gr Ov T, E 1 Vera Bitters 77 19 Sd, Gr Ov T, T 1 D. Brown 63 R 19 Sd, Gr Ov T, E 1 Roberta West 62 R 16 Sd, Gr Ov T, E 1 Bertha Weltmer 62 R 19 Sd, Gr Ov T, E 1 D. Brown 54 19 Sd, Gr Ov T, E 1 M. West 58 19 Sd, Gr Ov T, E 1 Go 84 R 19 Sd, Gr Ov T, E 1 Frank West 52 R 19 Sd, Gr Ov T, E 1 Gharles Collins 64 R 20 1 1 Ralph Dome 64 R 19 Sd, Gr Ov T, E 1	V-21ddc 22acb 22bcb 22cbb 22ccc	do H. J. Ford H. O. Jones Charles Collins do	85 45 45 45 84 84 84 84 84 84 84 84 84 84 84 84 84	16 19 20 16	чO		\$\$\$\$\$	r t.t.t.t. mmmm	нннн	24.6 23.7 20.0 20.3 27.0	77.00 27.00 27.00 20.00	2,02,02,02,02,02,02,02,02,03,02,03,03,03,03,03,03,03,03,03,03,03,03,03,	500 R 500 R
do 63 R 19 Sd, Gr Qv T, E I Robert Trout 65 R 19 Sd, Gr Qv T, T I Robert Trout 60 18 Sd, Gr Qv T, E I do 19 Sd, Gr Qv T, E I M. West 52 R 19 Sd, Gr Qv T, E I Frank West 52 R 19 Sd, Gr Qv T, E I Charles Collins 64 20 I Sd, Gr Qv T, E I Ralph Dome 64 R 19 Sd, Gr Qv T, E I	22dba 22ddb 23cbc 23ccd 24bcd	L. J. Ford Frank Kershner John Weitzel Vera Bitters D. Brown	70 78 R 75 77 81 R	19 19 19			<u> </u>	HEEHH HEEHH	нчнч	19.8 21.5 22.0 23.8 23.0	44469 60000 90000	2,025 2,026 2,027 2,027 2,021	450 R
M. West 58 19 Sd, Gr Qv T, T I do 84 R 19 Sd, Gr Ov T, E I Frank West 52 R 19 Sd, Gr Ov T, E I Charles Collins 64 20 I Sd, Gr Ov T, E I Ralph Dome 64 R 19 Sd, Gr Ov T I I	24cdb 24dbc 25aba 25acb 25cbd	do Bertha Weltmer Robert Trout do D. Brown	63 R 62 R 60 R 54	19 19 16 18			\$\$\$\$\$	1, E 1, T 1, E 1, E	нннн	21.9 24.1 20.8 21.3 25.5	, w w w e w 6000000000000000000000000000000000000	2,018 2,019 2,014 2,014 2,021	500 600 850 R 900
	26aab 26add 26bbb 27aaa* 7-13dad	M. West do Frank West Charles Collins Ralph Dome	58 848 64 64 8	19 19 20 19	H		\$\$\$\$\$	HHHH HHHH	ппппп	18.9 21.4 24.0 24.6 35.7	600000 6000000000000000000000000000000	2,019 2,019 2,029 2,028	600 R 700 R 1,200 650 R

700 R 450 R 450 600 R	600 R	400 R	500 R	400 R		500 R 450 R	400 600 R		290		
2,052 2,051 2,056 2,064 2,074	2,078 2,080	2,084 2,077 2,073	2,076 2,079 2,074 0,078	2,066	2,066 2,070	2,063 2,061 2,065	2,049 2,052	1,982 1,975	2,020	2,205 2,145 9,95	2,235
09-9 09-9 09-9	09-9 09-9	09-9 09-9 09-9	6-60 6-60 10-60	09-9	09-9 09-9	6-60 6-60 10-59	09-9	2-60	5-60		
26.7 24.9 26.0 25.4 29.7	25.3 26.4	32.4 26.8 26.5	26.4 28.9 0.0 0.0	25.7	24.9 24.3	27.5 27.0 31.8	24.2	22.1	23.6		
нанны	нн	ннн	I PS	· H		SHHS	пп ^с	U,C V,S	D, S	JQ SSSS	s,2
T T T, NG T, LPG	ΕĽ	C, NG T	HH,Z	4H	C, NG	C, LPG N	T, LPG	Cy, E, W T, W	Cy, E T, T	ÇÇÇ, E	Cy, E. W
<u> </u>	òò		8888	ŠŠ	òċ	, \$\delta\de	000	ov, Kd Qt, Kd	Kd Qt, Kd	KKYG KKKY	Kq
SSS SS SS SSS SS	Sd, Gr Sd. Gr	Sd, Gr Sd, Gr Sd, Gr	Sd, Gr Sd, Gr Sd, Gr	Sd, Gr	Sd, Gr	Sd, Gr Sd, Gr Sd, Gr	Sd, Gr Sd, Gr	ss Ss, Sd Ss, Sd	Ss. Gr	8 8 8 5	SS SS
					m m	: v	S		Ŋ		S
19 26 16 12	18	72	16 18	12	25 E	248 248	30 18	6 14	3	9	w
61 61 R 63 R 53	47 50 B	228 R	55 52 R 47	43	47	49 51 R 50	52 53 R	214 K 80 140 R	185 R 140 R	290 R 284 R	410 R
do do John Georg Clyde Bryant Harry Jones	Fred Pabst do	L. C. Young V. Huxol W. R. Olson	Mrs. J. G. Bott do City of Alexander	Olikhown Clyde Bryant	G. W. Geckles	John C. Seltman George Miller G. Wilson	George Kershner Lynn Fox	kenneth Almquist Herman Hemken Wm. F. Werhan	Les Brannon Ed Holopirek	Martin nemken A. Eilts B. Franker	Claude Huddleston
13dbc 13dcc 14ccb* 15ccb* 17dcc	19aab 19aac	19acc 19daa 20acc	20bcd 20cbc 20dca*	2000c 21acc	21caa 21cho	22cbb 22dba 23cda	24aab 24abc	36abb* 19-16W-23ddb* 36dab	19-17W-27add* 19-18W- 3aca*	19-19W-24ccb* 19-20W- 4bbb*	26baa*

Well-numbering system described in text.
Asterisk following well number indicates analysis of water is given in table 10.
R. reported; E, estimated.
G galvanized; I, black iron; R. rock; S, steel.
GC, gravel; Sd, sand; Sh, shale; Ss, sandstone; St, silt.
C c. Carlile Shale; Kd, Dakota Formation; Qt, terrace deposits; Qu, undifferentiated Pleistocene deposits; Qv, valley-fill deposits.
C centrifugal; Cy, cylinder; J, jet; M, none; S, submersible; T, turbine.
E, electric; H, hand; LPG, liquefied petroleum gas; NG, natural gas; T, tractor; W, wind.
D, domestic; I, irrigation; PS, public supply; S, stock.

Thickness, Depth,

LOGS OF TEST HOLES

The logs of 6 test holes drilled or augered by the U.S. Geological Survey and the State Geological Survey of Kansas and one test hole drilled by Dal Wells are given on the following pages. Logs of additional test holes are on file at the state and U.S. Geological Survey offices, Lawrence, Kans., and may be examined there.

17-19W-21aaa.—Sample log of test hole in the NE¼NE¼NE¼ sec. 21, T.17 S., R.19 W., in trail into field about 13 feet east and 7 feet south of stone cornerpost in NE corner of section; augered August 2, 1960. Altitude of land surface, 2,222.6 feet. Depth to water, 23.5 feet.

	Thickness,	
QUATERNARY SYSTEM	feet	feet
PLIOCENE SERIES		
Ogallala Formation		
Soil	3.5	3.5
Silt with caliche nodules	2.5	6
Silt and clay; some sand	2.5	8.5
Sand, very fine to coarse	10	18.5
Sand and gravel, fine to coarse	9.5	28
CRETACEOUS SYSTEM		
Upper Cretaceous Series		
Colorado Group		
Carlile Shale		
Shale, weathered	.5	28.5
18-16W-24bcc.—Sample log of test hole in the	SW4SW	4NW4

sec. 24, T.18 S., R.16 W., about 60 yards east of Otis blacktop in north track of trail going east just south of bridge over Walnut Creek; drilled September 28, 1960. Altitude of land surface, 1,930.7 feet. Depth to water, 17.1 feet. Thickness, Depth,

QUATERNARY SYSTEM	†eet	feet
PLEISTOCENE SERIES		
VALLEY-FILL DEPOSITS		
Roadfill and soil	10	10
Silt and clay, brown	19	29
Clay, silty grayish-white; some sand,	10	20
very fine; some mollusk shells	4	33
Sand, fine to coarse	$rac{4}{7}$	40
Sand, fine to coarse; gravel, fine; some	•	10
clay	10	50
Sand, fine to coarse; gravel, fine to me-		
dium	5	55
Sand, fine to coarse; strip of clay at 75		
feet; few mollusk shells	35	90
Sand, very fine; some clay	8	98
Sand, fine to coarse; gravel, fine	17	115
Sand, fine; some gravel	5	120
CRETACEOUS SYSTEM		
LOWER AND UPPER(?) CRETACEOUS SERIES		
DAKOTA FORMATION		
Sand, fine to coarse (cemented with iron		
oxide); some clay, red and gray	10	130
Sand, fine (cemented with iron oxide)	20	150
10 1877 011		

18-17W-31bbc.—Sample log of test hole in the SW¼NW¼NW¼ sec. 31, T.18 S., R.17 W., in east road shoulder about 21 yards south of sign for narrow bridge across Walnut Creek tributary or about 80 yards north of bridge; drilled September 22 to 26, 1960. Altitude of land surface, 2,000.1 feet. Depth to water, 21.0 feet.

	et
QUATERNARY SYSTEM	
Pleistocene Series	
Valley-fill deposits	
Roadfill4	4
Silt and clay, tan to brown 16 2	Ô
Clay and silt, brown to black 8 9	8
Clay, silty, bluish-gray; some sand, very	٠.
fine	0
Clay, silty, buish-gray; sand, coarse,	

quartz; many fossil mollusks	2	32
quartz; gravel, fine limestone	6	38
Clay, gray; sand, fine, quartz; many fos-	4	42
Sand, coarse, quartz; gravel, fine to medium, limestone, angular, some quartz, rounded	8	50
Sand, very fine to coarse, quartz; gravel, fine to coarse, limestone, rounded.	O	50
some arkosic material; many fossil mollusks; some clay, brown to black	25	75
Sand, fine to coarse; gravel, fine to coarse composed of limestone; clay,		
silty, grayCRETACEOUS SYSTEM	17	92
Lower and Upper(?) Cretaceous Series Dakota Formation		
Clay, silty, sandy, gray to ochre	18	110
18-18W-28age - Sample log of test hale in the	CXX71/N	TIC1/NTIC1/

18-18W-28aac.—Sample log of test hole in the SW¼NE¼NE¾ sec. 28, T.18 S., R.18 W., two blocks south of Kans. Hwy. 96 and two blocks west of U.S. Hwy. 182, 14 yards north and 144 yards west of SW corner of vacant lot NW of school in Rush Center. Site of Rush Center city well; drilled by Dal Wells for Rush Center September 15, 1960. Altitude of land surface, 1,996.0 feet. Depth of water, 20.0 feet.

QUATERNARY SYSTEM	†eet	ţeet
PLEISTOCENE SERIES		
VALLEY-FILL DEPOSITS		
C ·1	0	
	3	3
Silt and clay, brown	18	21
Silt and clay, brown; some sand, fine	3 2 6 3 1 3	24
Sand, fine; some silt, brown	2	26
Sand, fine to coarse	6	32
Sand, fine to coarse; gravel, fine	3	35
Clay and sand	1	36
Sand, fine to coarse; gravel, fine	3	39
Sand, fine to coarse; gravel, fine; clay		
streaks	15	54
Sand, fine to coarse; gravel, fine to me-		
dium	11	65
Clay streak and sand	ī	66
Sand, fine to coarse; gravel, fine to me-	_	00
dium	6	72
Sand, very fine; clay, reddish-brown	11	83
Sand, fine to coarse; gravel, fine to me-	11	00
	6	89
Sand with clay	3	
Sand fine to coorse, graved fine to	ა	92
Sand, fine to coarse; gravel, fine to me-	0	101
	9	101
CRETACEOUS SYSTEM		
Lower and Upper(?) Cretaceous Series		
Dakota Formation		
Shale, red; some sand	3	104
18-20W-19add.—Sample log of test hole in th	e SE4S	E4NE

sec. 19, T.18 S., R.20 W., about 100 yards north of half-mile line in west barpit; drilled September 22, 1960. Altitude of land surface 2, 2075 3 fast P. R. Henry 1960.

land surface, 2,075.3 feet. Depth to water 27.3	feet.	
QUATERNARY SYSTEM	Thickness, feet	Depth, feet
PLEISTOCENE SERIES		
Valley-fill deposits		
Silt, light-brown	10	10
Silt and clay, light-brown; caliche zone		
at 13 feet about 0.2 foot thick	10	20
Silt and clay, tan to light-brown	8	28
Sand, very fine to fine	7	35
Sand, very fine to fine with some clay		
bluish-black	8	43
Sand, fine to coarse; gravel, fine to me-		
dium; some clay	4	47
CRETACEOUS SYSTEM		
UPPER CRETACEOUS SERIES		
Colorado Group		
Graneros Shale		
Shale, platy, hard, blackish-gray	3	50
, , , , , , , , , , , , , , , , ,	_	-0

19-16W-23dad.—Sample log of test hole in the SE¼NE¼SE½ sec. 23, T.19 S., R.16 W., in west road shoulder directly east of first line pole north of drive into farmstead; drilled September 29, 1960. Altitude of land surface, 1,980.8 feet. Depth to water, 28.9 feet.

Thickness, Depth,

	feet	
QUATERNARY SYSTEM	,	,
Pleistocene Series		
Valley-fill deposits		
Roadfill and silt, reddish-brown	10	10
Silt and clay, reddish-brown	40	50
Silt and clay, tan, soft, with caliche	6	56
Silt and clay, reddish-brown	4	60
Clay and silt, soft to hard, tannish-red to		
reddish-brown, with caliche	16	76
Clay and silt, black	4	80
Clay, black; gravel, fine to medium,		
limestone, angular; many mollusk		
shells	10	90
Clay, soft, silty, tan	5	95
Gravel, fine, limestone, angular	6	101
CRETACEOUS SYSTEM		
LOWER AND UPPER(?) CRETACEOUS SERIES		
DAKOTA FORMATION		
Clay, black, gray, and green	9	110
Rock chips; sand, fine; clay, varigated		
red	10	120
Sand, fine, limonitic cement; some clay	10	130
•		

19-18W-33aba.—Sample log of test hole in the NE¾NW¾NE¾ sec. 33, T.18 S., R.18 W., in the south road shoulder about 500 yards west of U.S. Hwy. 183; augered June 27, 1960. Altitude of land surface, 2,214.1 feet. Depth to water, 34.0 feet.

	Thickness, feet	Depth, feet
QUATERNARY SYSTEM	,,,,,,	,
PLIOCENE SERIES		
Ogallala Formation		
Roadfill	. 3.5 . 5 . 8	3.5
Silt, some sand	. 5	8.5
Sand, very fine, "flour like," pink		16.5
Sand, fine, pink		20.5
Sand, fine, grayish-pink	. 7	27.5
Sand, fine to medium, gray	. 10	37.5
Clay, tan; some sand	. 6	43.5
CRETACEOUS SYSTEM		
Upper Cretaceous Series		
Colorado Group		
Carlile Shale		
Shale, dark-grey	.5	44

GLOSSARY OF TERMS

Many of the county reports of ground-water investigations in Kansas contain summaries of the principles of occurrence and movement of ground water as discussed by Meinzer (1923a) and Moore and others (1940); the reader is referred to Lane (1960, p. 25-39), Bayne (1960, p. 25-30), or Hodson and Wahl (1960, p. 28-32) for a discussion of the subject. Most of the following definitions of technical terms commonly used in a discussion of ground water are from Leonard and Berry (1961, p. 58-61), as adapted from Meinzer (1923b); however, some are from Lohman and others (1972).

Aquifer.—A rock formation, bed, or zone that contains water that is available to wells. An aquifer is sometimes referred to as a water-bearing rock, or water-bearing bed.

Artesian water.—Ground water under sufficient hydraulic pressure to rise above the level at which

the water-bearing rock is tapped in drilling a well. Equivalent to confined ground water. The pressure is sometimes called artesian pressure and the rock containing artesian water is an artesian aquifer or confined aquifer.

Colluvial.—Of or relating to gravitational processes. Colluvial deposits result from action of gravity pulling loose material downhill.

Confining bed.—A less permeable rock layer that overlies or underlies an aquifer and retards vertical movement of water.

Eolian.—Of or relating to wind or wind action. Eolian deposits result from deposition of wind-carried sediment.

Evapotranspiration.—The combined total water evaporated by heat energy and transpired by plants into the atmosphere.

Fluvial.—Of or relating to streams or running water. Fluvial deposits result from deposition of water-carried sediment.

Gaining stream.—A stream or reach of a stream whose flow is being increased by inflow of ground water. Replaces the term "effluent stream."

Ground water.—Water in the saturated zone or water below the water table.

Hydraulic conductivity.—A measure of the rate of flow of water through an aquifer, which is dependent primarily on the nature of the interstices within the aquifer. Expressed in units of length per units of time that are consistent and suitable to the problem involved. Replaces the term "field coefficient of permeability." To convert a value of hydraulic conductivity to a value for coefficient of permeability, multiply by 7.48.

Hydraulic gradient.—Gradient of the water table measured in the direction of the greatest slope, generally expressed in feet per mile.

Impermeable rock.—An impervious rock, through which movement of water under common subsurface pressure differentials is negligible.

Inflow.—Movement of ground water into an area in response to a hydraulic gradient.

Interstice.—An opening or void in a rock. Interstices may be filled with air, gas, oil, water, or some other material. The interstices in an aquifer are filled with water.

Losing stream.—A stream or reach of a stream that is losing water to the ground. Replaces the term "influent stream."

Outflow.—Movement of ground water from an area in response to a hydraulic gradient.

Percolate.—The movement of water through soil and rock to the saturated zone.

Permeable rock.—Pervious rock, or a rock that has a texture permitting water to move through it readily under ordinary pressure differentials.

Permeability.—The capacity of water-bearing rock to transmit water, which is related to the size and interconnection of interstices. The field coefficient of permeability, is expressed as the rate of flow of water at the prevailing temperature, in gallons per day, through each mile of width and for each foot of thickness of an aquifer under a hydraulic gradient of 1 foot per mile; replaced by the term "hydraulic conductivity." To convert a value for field coefficient of permeability to the equivalent value of hydraulic conductivity, multiply by 0.134.

Porosity.—The porosity of a rock is its property of containing openings or interstices. Quantitatively, the porosity of a rock is the ratio (usually expressed as a percentage) of the volume of openings in the rock to the total volume of the rock.

Recharge.—The process by which water is absorbed and added to the saturated zone. Also used to designate the quantity of water added to the ground-water reservoir.

Runoff.—The discharge of water through surface streams. It includes both surface-water runoff and ground-water runoff. Also used to designate the quantity of water discharged as runoff.

Saturated zone.—The zone of porous rocks saturated with water. Ground water is contained in this zone.

Storage.—Water stored in openings in the saturated zone is said to be in storage. Discharge of water from an aquifer not replaced by recharge is said to be from storage.

Storage coefficient.—The volume of water released from or taken into storage per unit surface area of an aquifer per unit change in the component of head normal to that surface.

Transmissibility.—The transmissibility of a rock is its capacity to transmit water under pressure. The coefficient of transmissibility is the field coefficient of permeability multiplied by the saturated thickness, in feet, of the aquifer; replaced by the term "transmissivity." To convert a value of transmissivity to a value for coefficient of transmissibility, multiply by 7.48.

Transmissivity.—The rate at which water is transmitted through a unit width of an aquifer under a unit hydraulic gradient. Expressed in units of length squared per units of time. Replaces the term "coefficient of transmissibility." To convert a value for coefficient of transmissibility to an equivalent value of transmissivity, multiply by 0.134.

Water table.—The upper surface of the saturated zone where the pressure is atmospheric. The water table is not a plane surface, but has irregularities much like the land surface.

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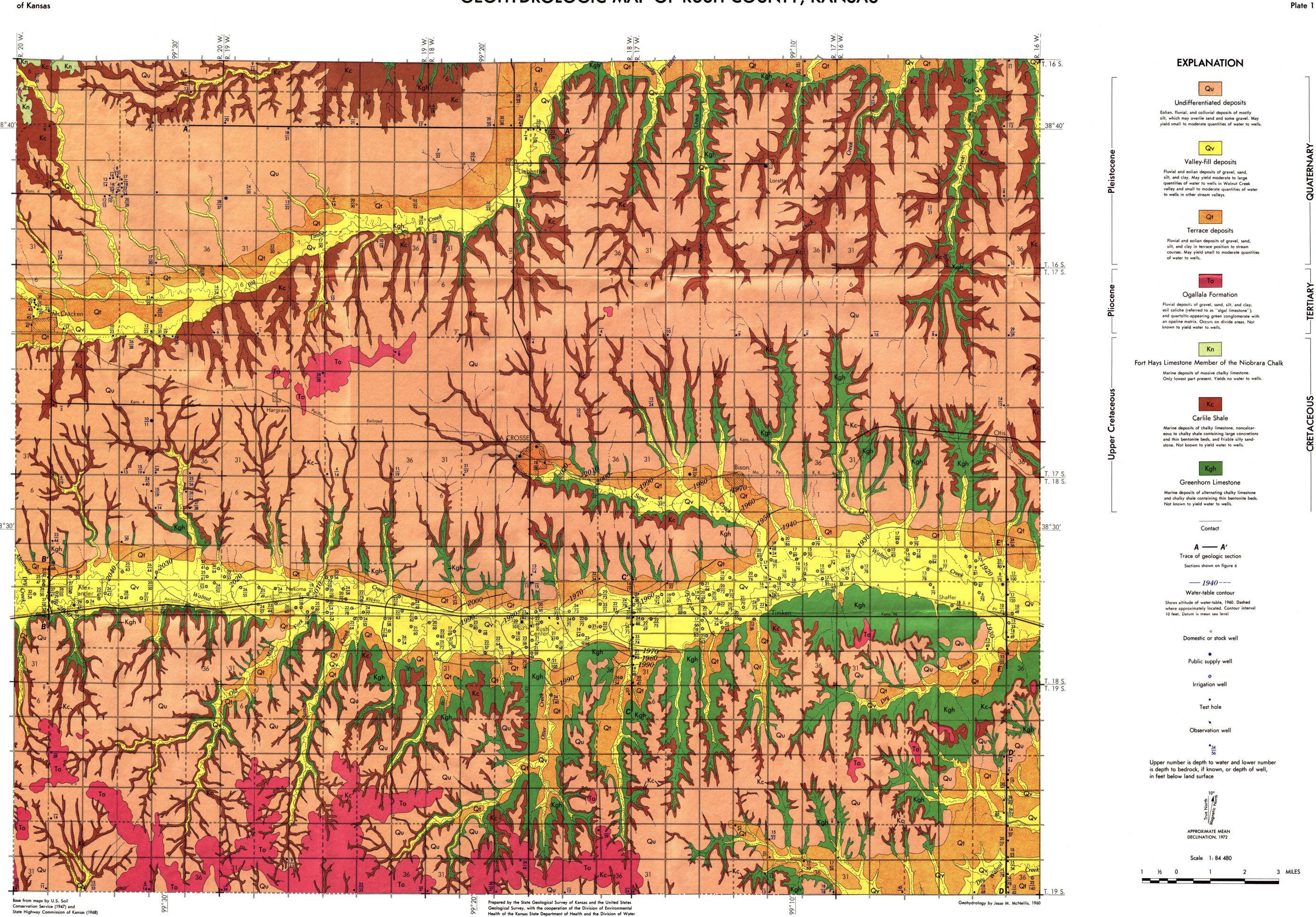
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showing configuration of bedrock surface, saturated thickness (1960), and decline in water level (1960-7

